

## Chemistry Semester 1 Final Exam Review

1. What is a definition of Chemistry?  
**Study of matter**
  2. Matter includes all of the following EXCEPT:  
a. air      **b. light**      c. smoke      d. water vapor
  3. What is matter?  
**Something that has mass and takes up space (volume)**
  4. A physical change occurs when a  
a. peach spoils      c. copper bowl tarnishes  
b. bracelet turns your wrist green      **d. glue gun melts a glue stick**
  5. The particles in a solid are  
**a. packed closely together**      c. constantly in motion  
b. very far apart      d. able to slide past one another
- Of the following groups, 1, 2, 14, 15, 17, 18:
6. Group(s) **1, 2** contains only metals
  7. Group (s) **18** is called a Noble gases.
  8. Group(s) **17** has 7 electrons on the valence shell
  9. Group(s) **15** needs to gain 3 electrons to fill a stable octet
  10. Group(s) **2** has a 2+ oxidation state
  11. Group(s) **2** has an electron dot notation of  $\bullet X \bullet$
  12. Using the periodic table, an element similar to carbon would be **14**
  13. Based on their location on the periodic table, you could infer that **18/noble gases** are very unreactive
  14. The horizontal row on the periodic table is called a(n) **period**
  15. The vertical column of blocks on the periodic table is called a(n) **group/family**
  16. The elements on the zigzag line in the periodic table are **metalloids**
  17. Which is NOT a property of metal?  
a. malleability      c. **unreactivity**  
b. ability to lose electrons      d. ability to conduct heat
  18. A volume of 1 cc ( $\text{cm}^3$ ) is the same as **1** mL.
  19. To determine density, the quantities that must be measured are **volume** and **mass**
  20. The relationship between the mass  $m$  of a material, its volume  $V$ , and its density  $D$  is  
a.  $V = mD$       b.  $V/m = D$       **c.  $DF = m$**       d.  $D + V = m$
  28. The density of aluminum is  $2.70 \text{ g/cm}^3$ . The volume of a solid piece of Aluminum is  $1.50 \text{ cm}^3$ . Find the mass **4.05 grams**
  21. The density of pure diamond is  $3.5 \text{ g/cm}^3$ . The mass of a diamond is 0.25 g. Find the volume= **0.71  $\text{cm}^3$**
  22. What statement about density is true?  
a. two samples of a pure substance may have different densities if they are different sizes  
b. the density of a sample depends on its location on Earth  
c. a cylinder is always used to measure the volume  
**d. density is a physical property that remains constant for a pure substance**
  23. The number of grams equal to 0.5 kg is **500 g**
  24. In a graph the **independent** variable is placed on the x axis and the **dependant** variable is placed on the y axis.
  25. The variable that you change in the lab is the **independent** variable. The variable that responds to the change is called the **dependant** variable. All other things do not change and are called **constants**
  26. At first Dalton thought the atom was solid, spherical, **indivisible**, then Rutherford discovered the **nucleus** of the atom, then the subatomic particles **protons**, **neutrons**, and **electrons** were discovered
  27. The positively charged particle is the **proton** found in the **nucleus** of the atom
  28. The subatomic particle that has about the same mass as the proton, but with no electrical charge, is called a **neutron** and is found in the **nucleus** of the atom
  29. The **electron** is negatively charged and is found in the **electron cloud/outer shell** of the atom
  30. **Bohr** is the person credited with placing electrons on levels and **Schrodinger** established the electron cloud theory.
  31. The forces that hold the particles together in the nucleus are **columbic /electrostatic** forces
  32. Isotopes are atoms of the same element that **have different number of neutrons**
  33. All isotopes of hydrogen contain **1 proton & 1 electron**
  34. What is the atomic number for Nitrogen? **7** To what subatomic particle does this correspond? **# of protons**
  35. Zinc -66 has **30** protons and **36** neutrons
  36. What is the electron configuration for sodium?  **$[\text{Ne}]3s^1$**  Chlorine?  **$[\text{Ne}]3s^23p^5$**  Neon?  **$[\text{He}]2s^22p^6$**
  37. What is the dot notation for magnesium? Bromine? Argon?

# Chemistry 1 Honors Final Exam Study Guide

**Peipei Pang**



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Student-Centered Learning by Design Jacquelyn Whiting, 2021-03-15 Current research is pushing schools to adopt more student centered approaches to the classroom experience and educators librarians and classroom teachers alike are being challenged to revise their curricula and instruction to be student centered personalized and differentiated This book empowers librarians teachers and administrators to be empathic problem solvers and decision makers By reframing the challenges that members of a learning community face as opportunities to better meet teaching and learning needs readers will find that adoption of a mindset focused on users namely design thinking elevates and creates opportunities for innovating pedagogy Moreover it can enhance school culture as well as build channels of communication among various stakeholders in schools and districts When educators of any subject or discipline apply design thinking skills to their curriculum implementation authentic student centered learning experiences become the core of the learning experience The case studies shared in this book provide examples of student centered approaches being used in elementary middle and high schools so that readers have many models on which to base their work and from which to build confidence in shifting their pedagogy to keep the student at the center of teaching and learning decisions *Teaching Innovation in University Education: Case Studies and Main Practices* Saura, Jose Ramon, 2022-06-17 In the last decade the development of new technologies has made innovation a fundamental pillar of education Teaching innovation includes the evolution of both teaching and learning models to drive improvements in educational methodologies Teaching innovation is a pioneer in the understanding and comprehension of the different teaching methodologies and models developed in the academic area Teaching innovation is a process that seeks validation in the academic and teaching communities at universities in order to promote the improvement and its practices and uses in the future characterized by digital development and data based methods Teaching Innovation in University Education Case Studies and Main Practices features the major practices and case studies of teaching innovation developed in recent years at universities It is a source on study cases focused on teaching innovation methodologies as well as on the identification of new technologies that will help the development of initiatives and practices focused on teaching innovation at higher education institutions Covering topics such as didactic strategics service learning and technology based gamification this premier reference source is an indispensable resource for pre service teachers lecturers students faculty administrators libraries entrepreneurs researchers and academicians **Lovejoy's Prep School Guide** Clarence Earle Lovejoy, 1974 *Chemistry for Higher Education* Jan H. Apotheker, 2018-08-21 Current publication gives hands on recommendations how to develop a successful course in either the bachelor or the master of chemistry The author discusses different ways of course building such as lectures workshops seminars and labs explains how to identify potential improvements for the next run of the class and elucidates the tools to create an efficient learning environment that helps students to understand the nature of chemistry *Guide to American Graduate Schools* Harold R.

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Dziuban,Anthony G. Picciano,2023-08-25 Data Analytics and Adaptive Learning offers new insights into the use of emerging data analysis and adaptive techniques in multiple learning settings In recent years both analytics and adaptive learning have helped educators become more responsive to learners in virtual blended and personalized environments This set of rich illuminating international studies spans quantitative qualitative and mixed methods research in higher education K 12 and adult continuing education contexts By exploring the issues of definition and pedagogical practice that permeate teaching and learning and concluding with recommendations for the future research and practice necessary to support educators at all levels this book will prepare researchers developers and graduate students of instructional technology to produce evidence for the benefits and challenges of data driven learning **The Next Five Years** Allan A. Glatthorn,Carl J.

Manone,1965 *EMRS PGT Chemistry Exam Book (English Edition) - Eklavya Model Residential School Post Graduate Teacher - 10 Practice Tests (1500 Solved Questions)* EduGorilla Prep Experts,2023-10-01 Best Selling Book in English Edition for EMRS PGT Post Graduate Teacher Chemistry Exam with objective type questions as per the latest syllabus Compare your performance with other students using Smart Answer Sheets in EduGorilla s EMRS PGT Post Graduate Teacher Chemistry Exam Practice Kit EMRS PGT Post Graduate Teacher Chemistry Exam Preparation Kit comes with 10 Practice Tests with the best quality content Increase your chances of selection by 16X EMRS PGT Post Graduate Teacher Chemistry Exam Prep Kit comes with well structured and 100% detailed solutions for all the questions Clear exam with good grades using thoroughly Researched Content by experts *Guide to American Graduate Schools* ,1982 **Resources in Education** ,1998

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United States Air Force Academy United States Air Force Academy, *Iowa Orientation* University of Iowa,2007

*REA's Authoritative Guide to Medical & Dental Schools* Research and Education Association, 1996-12-01 An excellent source book for those who are beginning the medical or dental school application process Included are profiles on every U S And Canadian medical and dental school as well as information on select foreign medical schools Also included are sections on osteopathic schools chiropractic schools and podiatric schools Important information is also included on undergraduate preparation the application process financial aid and graduation requirements      **Compact Guide to Colleges** Tessa Krailing, Barron's Publishing, Barron's Educational Series, 1992-07 This smaller version of Barron's definitive Profiles of American Colleges presents detailed descriptions of more than 400 accredited four year schools that fall mainly within the top three categories of Barron's exclusive academic competitiveness scale Updated with the latest facts and figures each of the Compact Guide's college profiles includes information on admission requirements academic programs tuition and fees available financial aid library and computer facilities student faculty ratios and much more

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