

Cations \ Anions	+	-	chloride	oxide	iodide	hydride
			$\text{Cl}^-$	$\text{O}^{2-}$	$\text{I}^-$	$\text{H}^-$
Sodium	$\text{Na}^+$		$\text{NaCl}$	$\text{Na}_2\text{O}$	$\text{NaI}$	$\text{NaH}$
Potassium	$\text{K}^+$		$\text{KCl}$	$\text{K}_2\text{O}$	$\text{KI}$	$\text{KH}$
Magnesium	$\text{Mg}^{2+}$		$\text{MgCl}_2$	$\text{MgO}$	$\text{MgI}_2$	$\text{MgH}_2$
Calcium	$\text{Ca}^{2+}$		$\text{CaCl}_2$	$\text{CaO}$	$\text{CaI}_2$	$\text{CaH}_2$
Copper(II)	$\text{Cu}^{2+}$		$\text{CuCl}_2$	$\text{CuO}$	$\text{CuI}_2$	$\text{CuH}_2$
Iron(II)	$\text{Fe}^{2+}$		$\text{FeCl}_2$	$\text{FeO}$	$\text{FeI}_2$	$\text{FeH}_2$
Iron(III)	$\text{Fe}^{3+}$		$\text{FeCl}_3$	$\text{Fe}_2\text{O}_3$	$\text{FeI}_3$	$\text{FeH}_3$
Silver	$\text{Ag}^+$		$\text{AgCl}$	$\text{Ag}_2\text{O}$	$\text{AgI}$	$\text{AgH}$
Zinc	$\text{Zn}^{2+}$		$\text{ZnCl}_2$	$\text{ZnO}$	$\text{ZnI}_2$	$\text{ZnH}_2$
Aluminum	$\text{Al}^{3+}$		$\text{AlCl}_3$	$\text{Al}_2\text{O}_3$	$\text{AlI}_3$	$\text{AlH}_3$

## Set 2

Cations \ Anions	+	-	bromide	oxide	fluoride	astatide
			$\text{Br}^-$	$\text{O}^{2-}$	$\text{F}^-$	$\text{At}^-$
Lithium	$\text{Li}^+$		$\text{LiBr}$	$\text{Li}_2\text{O}$	$\text{LiF}$	$\text{LiAt}$
Barium	$\text{Ba}^{2+}$		$\text{BaBr}_2$	$\text{BaO}$	$\text{BaF}_2$	$\text{BaAt}_2$
Cesium	$\text{Cs}^+$		$\text{CsBr}$	$\text{Cs}_2\text{O}$	$\text{CsF}$	$\text{CsAt}$
Strontium	$\text{Sr}^{2+}$		$\text{SrBr}_2$	$\text{SrO}$	$\text{SrF}_2$	$\text{SrAt}_2$
Copper(I)	$\text{Cu}^+$		$\text{CuBr}$	$\text{Cu}_2\text{O}$	$\text{CuF}$	$\text{CuAt}$
Copper(II)	$\text{Cu}^{2+}$		$\text{CuBr}_2$	$\text{CuO}$	$\text{CuF}_2$	$\text{CuAt}_2$
Lead(II)	$\text{Pb}^{2+}$		$\text{PbBr}_2$	$\text{PbO}$	$\text{PbF}_2$	$\text{PbAt}_2$
Lead(IV)	$\text{Pb}^{4+}$		$\text{PbBr}_4$	$\text{PbO}_2$	$\text{PbF}_4$	$\text{PbAt}_4$
Gallium	$\text{Ga}^{3+}$		$\text{GaBr}_3$	$\text{Ga}_2\text{O}_3$	$\text{GaF}_3$	$\text{GaAt}_3$
Nickel(II)	$\text{Ni}^{2+}$		$\text{NiBr}_2$	$\text{NiO}$	$\text{NiF}_2$	$\text{NiAt}_2$

# Chemical Formula Writing In Table Answer Key

**Morris Hein, Susan Arena**



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*The IT in Secondary Science Book* Roger Frost,1994

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**Chemistry Textbook for College and University USA** Ibrahim Sikder, 2023-06-04 Chemistry Textbook USA

**Chemistry for OCR A for Double Award** David Lees, John Payne, 2001 This series is for schools following OCR A double or separate award for GCSE science The resources offer preparation for the OCR exams with teacher support to minimise time spent on administration The teacher's resources are available on CD ROM in a fully customizable format

**Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van't Hoff factor and applications of radioactivity

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**Learning Chemistry 6 Solution Book (Year 2023-24)**, 2024-01-02 Chemistry 'O' Level

Rex M. Heyworth, 2007 *Introductory Chemistry* Charles H. Corwin, 2005 For one semester courses in Basic Chemistry Introduction to Chemistry and Preparatory Chemistry and the first term of Allied Health Chemistry This text is carefully crafted to help students learn chemical skills and concepts more effectively Corwin covers math and problem solving early in the text he builds student confidence and skills through innovative problem solving pedagogy and technology formulated to meet student needs

International Critical Tables of Numerical Data, Physics, Chemistry and Technology National Research Council (U.S.), Callie Hull, 1928

**Scientifica**, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

*Foundations of College Chemistry, Alternate* Morris Hein, Susan Arena, 2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They'll learn how to apply concepts with the help of worked out examples In addition

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