

Chemistry Bonding MCQ (From 2015 Prelim Papers)

Q1

In which of the following pairs does the first species have a larger bond angle than the second?

- A SO_2 , PI_3
- B SO_3^{2-} , SO_4^{2-}
- C H_2S , NH_3
- D H_2O , H_3O^+


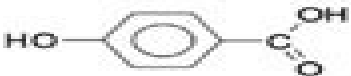
Q2

Which of the following statements is correct?

- A BCl_3 and BrCl_3 have the same molecular shape.
- B Cyclohexene and propanone are non-polar liquids.
- C LiAlH_4 and NaBF_4 contain dative covalent bonds.
- D XeF_4 and $\text{H}_2\text{C}=\text{C}=\text{CH}_2$ are planar molecules.


Q3

In which of the following pairs does the first substance have a higher melting point than the second?

	First substance	Second substance
A	$\text{CH}_3\text{CH}_2\text{OCH}_3$	$\text{CH}_3\text{CH}_2\text{NHCH}_3$
B	$\text{CH}_3(\text{CH}_2)_4\text{CH}_3$	$(\text{CH}_3)_2\text{CHCH}(\text{CH}_3)_2$
C	RbCl	KCl
D		

Q4

Which of the following is a planar molecule?

- A $\text{C}_2\text{F}_4^{2+}$
- B $\text{CH}_2=\text{C}=\text{CH}_2$
- C 
- D $\text{CH}_2=\text{CH}-\text{CHO}$

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