#### CHAPTER 1

#### 1.1. For the thermal cracking of ethane in a tubular reactor, the following data were obtained for the rate coefficient at different reference temperatures:

| T(*C)                | 702  | 725   | 734   | 754   | 773   | 789   | 803   | 810   | 827   | 837   |
|----------------------|------|-------|-------|-------|-------|-------|-------|-------|-------|-------|
| Bo(m <sup>-1</sup> ) | 0.15 | 0.273 | 0.333 | 0.595 | 0.923 | 1.492 | 2.138 | 2.718 | 4.137 | 4.665 |

#### Solution

The Arrhenius expression

$$k = A \exp\left(-\frac{E}{RT}\right)$$

is transformed logarithmically into:

$$ln\,k = ln\,A - \frac{E}{RT}$$

For each data point In k and 1/T is calculated:

| $x = 1/T \cdot 10^{\alpha}$ | $y = \ln k$ |
|-----------------------------|-------------|
| 1.025                       | -1.897      |
| 1.002                       | -1.298      |
| 0.993                       | -1.100      |
| 0.974                       | -0.519      |
| 0.956                       | -0.080      |
| 0.941                       | 0.400       |
| 0.929                       | 0.760       |
| 0.923                       | 1.000       |
| 0.909                       | 1.420       |
| 0.901                       | 1.540       |
|                             |             |

The slope and the intercept (In A) are calculated by linear regression:

$$-\frac{E}{R} = \frac{\sum xy - \frac{\sum x\sum y}{N}}{\sum x^2 - \frac{(\sum x)^2}{N}}$$

$$\ln A = \frac{\sum y - m \sum x}{N}$$

with x = 1/T and  $y = \ln k$ .

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# **Table of Contents Chemical Reactor Analysis Rawlings Solution Manual**

- 1. Understanding the eBook Chemical Reactor Analysis Rawlings Solution Manual
  - The Rise of Digital Reading Chemical Reactor Analysis Rawlings Solution Manual
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Chemical Reactor Analysis Rawlings Solution Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemical Reactor Analysis Rawlings Solution Manual
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chemical Reactor Analysis Rawlings Solution Manual
  - Personalized Recommendations
  - Chemical Reactor Analysis Rawlings Solution Manual User Reviews and Ratings
  - Chemical Reactor Analysis Rawlings Solution Manual and Bestseller Lists

- 5. Accessing Chemical Reactor Analysis Rawlings Solution Manual Free and Paid eBooks
  - Chemical Reactor Analysis Rawlings Solution Manual Public Domain eBooks
  - Chemical Reactor Analysis Rawlings Solution Manual eBook Subscription Services
  - Chemical Reactor Analysis Rawlings Solution Manual Budget-Friendly Options
- 6. Navigating Chemical Reactor Analysis Rawlings Solution Manual eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Chemical Reactor Analysis Rawlings Solution Manual Compatibility with Devices
  - Chemical Reactor Analysis Rawlings Solution Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemical Reactor Analysis Rawlings Solution Manual
  - Highlighting and Note-Taking Chemical Reactor Analysis Rawlings Solution Manual
  - Interactive Elements Chemical Reactor Analysis Rawlings Solution Manual
- 8. Staying Engaged with Chemical Reactor Analysis Rawlings Solution Manual
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemical Reactor Analysis Rawlings Solution Manual
- 9. Balancing eBooks and Physical Books Chemical Reactor Analysis Rawlings Solution Manual
  - Benefits of a Digital Library
  - $\circ\,$  Creating a Diverse Reading Collection Chemical Reactor Analysis Rawlings Solution Manual
- 10. Overcoming Reading Challenges
  - o Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Chemical Reactor Analysis Rawlings Solution Manual
  - Setting Reading Goals Chemical Reactor Analysis Rawlings Solution Manual
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemical Reactor Analysis Rawlings Solution Manual
  - Fact-Checking eBook Content of Chemical Reactor Analysis Rawlings Solution Manual
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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