

## Chemistry Worksheets Class 10 on Chapter 1 Chemical Reactions and Equations- Set 1

**Q-1:** Write the balanced chemical equations for each of the following reactions.

- Liquid hydrogen peroxide decomposes to form water and oxygen gas.
- Acetylene gas burns in oxygen to form carbon dioxide and water along with evolution of heat.
- When solid mercury(II) oxide is heated, liquid mercury and oxygen gas are produced.
- Marble dissolves in a hydrochloric acid to give calcium chloride, water and carbon dioxide.
- Copper sulphate on treatment with potassium iodide precipitates cuprous iodide, liberates iodine gas and also forms potassium sulphate.

**Q-2:** Fill in the blanks

- \_\_\_\_\_ is the magnetic oxide of iron.
- In reactivity series, manganese lies \_\_\_\_\_ zinc.
- To the reaction between  $\text{AgNO}_3$  solution and  $\text{NaCl}$  solution, the names given to this type of reactions are \_\_\_\_\_ and \_\_\_\_\_.
- Proteins decompose to form \_\_\_\_\_ in our body.
- If copper turnings are added to colourless  $\text{AgNO}_3$  solution, it turns \_\_\_\_\_ after sometime.

**Q-3:** Which of the following displacement reactions will not take place?

- $\text{Zn(s)} + \text{FeSO}_4\text{(aq)} \rightarrow \text{ZnSO}_4\text{(aq)} + \text{Fe(s)}$
- $\text{Fe(s)} + \text{CuSO}_4\text{(aq)} \rightarrow \text{FeSO}_4\text{(aq)} + \text{Cu(s)}$
- $\text{Cu(s)} + 2\text{AgNO}_3\text{(aq)} \rightarrow \text{Cu(NO}_3)_2\text{(aq)} + 2\text{Ag(s)}$
- $\text{Cu(s)} + \text{FeSO}_4\text{(aq)} \rightarrow \text{CuSO}_4\text{(aq)} + \text{Fe(s)}$

**Q-4:** Which of the following statements is false?

- Oxidation results in the loss of an electropositive element and the gain of an electronegative element.
- Reduction results in the loss of an electronegative element and the gain of an electropositive element.
- A species that gains electrons is referred to as a reducing agent.
- An oxidising agent is a type of specie that absorbs hydrogen atoms.

**Q-5:** Why are certain chemicals and medications kept in coloured bottles?

**Q-6:** Sakshi desired that her home be white washed. She went to the market and bought 20 kg of quick lime, which she dissolved in 60 litres of water. When she added lime to water, she noticed that the water began to boil even though it was not heated. Give an explanation for her observation. Name the product and write the corresponding chemical equation and word equation.

# Chemical Equations And Reactions Short Answers

**Kohli Nitin**



## Chemical Equations And Reactions Short Answers:

*Self-Help to CBSE Science Tenth Class Part 2 Chemistry (Solutions of Lakhmir Singh & Manjit Kaur)* Amar Bhutani, This book includes the answers to the questions given in the textbook CBSE Science Tenth Class Part 2 Chemistry published by S Chand Co and written by Lakhmir Singh and Manjit Kaur This book is based for latest syllabus **S.E.H. SCIENCE Class 10th** Himanshu Sharma, Priyanka Sharma, 2023-08-09 Educational Consultant Teacher STEP EDUCATION HUB Educational Institution S E H Science Class 10th *Longman Science Chemistry 10* Kohli Nitin, 2008-09 Science for Tenth Class Part 2 Chemistry Lakhmir Singh & Manjit Kaur, A series of books for Classes IX and X according to the CBSE syllabus and CCE Pattern *ARUN DEEP'S SELF-HELP TO ICSE CHEMISTRY CLASS 9 : 2025-26 EDITION (BASED ON LATEST ICSE SYLLABUS)* Amar Nath Bhutani, 2025-04-01 Self Help for I C S E Chemistry Class 9 has been meticulously crafted with the specific needs of 9th I C S E students in mind This comprehensive guide is designed to fully support students in preparing for exams effectively ensuring the attainment of higher grades The primary goal of this book is to assist every I C S E student in achieving their best possible grade by providing support throughout the course along with valuable advice on revision and exam preparation The material is presented in a clear and concise format enriched with ample practice questions KEY FEATURES Chapter At a Glance This section encompasses essential study material supported by definitions facts figures flow charts etc Solved Questions The condensed version is followed by solved questions and illustrative numerical s along with their answers solutions The book also includes answers to the questions given in the Concise Chemistry Class 9 textbook Competency based Questions Special questions based on the pattern of Olympiads and other competitions are included to familiarize students with the types of questions asked in competitions Experiments and Sample Question Papers To ensure completeness the book includes experiments and two sample question papers based on the exam pattern and syllabus Latest I C S E Specimen Question Paper The book concludes with the latest I C S E specimen question paper In summary Self Help for I C S E Chemistry for 9th class provides all the necessary material for examination success and will undoubtedly guide students on the path to success **Educart CBSE Class 10 Final Revision Book 2025 - Maths + Science + Social Science + English (2024-25)** Educart, 2024-12-18 What You Get Ch wise Important Q s Sample Papers Educart CBSE Class 10 Final Revision Book 2025 Maths Science Social Science English 2024 25 Strictly based on sample papers released by CBSE for 2025 exam preparation Includes ch wise important questions for each of the four subjects Includes unit wise quick revisions for each of the four subjects Practice questions from sample papers putting what you learnt to the test Why choose this book Best resource for structured and quick revision for the final board exams **Science For Tenth Class Part 2 Chemistry** LAKHMIR SINGH, A series of six books for Classes IX and X according to the CBSE syllabus *Chemistry* R P Manchanda, A text book on Chemistry *Gateway to Science — Chemistry for Class X* Dr. Chand Seth, Goyal Brothers Prakashan, 2020-01-01 Goyal Brothers Prakashan Study Material Based On NCERT

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