

Chapter 3: Chemical Bonding
Multiple choice questions (Part 1)

Multiple choice questions	Solution / reasons
1. Which of the following molecules has two lone-pairs of electrons? A. BrF_3 B. CHCl_3 C. H_2O_2 D. NH_3	
2. Which of the following molecules is linear? A. NO_2 B. H_2S C. HCN D. SO_3	
3. Which of the following best describe the change in bond angle in ammonia when the ammonium ion, NH_4^+ is formed? A. Decrease to approximately 92° B. Decrease to approximately 107° C. Increase to approximately 109° D. Increase to approximately 120°	
4. Which of the following species is tetrahedral in shape? A. XeF_4 B. ICl_4^- C. BrF_3 D. SO_4^{2-}	
5. Which of the following species is not planar? A. BF_3 B. I_3^- C. SiH_4 D. XeCl_4	
6. In which of the following species is the central atom sp^2 hybridised? A. $\text{CH}_2=\text{CH}_2$ B. PCl_3 C. CO_2 D. $\text{HC}\equiv\text{N}$	
7. The carbonate ion, CO_3^{2-} has three resonance structures as shown below <div style="display: flex; justify-content: space-around; align-items: center;"> $\left[\begin{array}{c} \text{:}\ddot{\text{O}}\text{:} \\ \parallel \\ \text{O}=\text{C}-\ddot{\text{O}}\text{:} \\ \parallel \\ \text{:}\ddot{\text{O}}\text{:} \end{array} \right]^{2-}$ $\left[\begin{array}{c} \text{:}\ddot{\text{O}}\text{:} \\ \parallel \\ \text{:}\ddot{\text{O}}-\text{C}=\ddot{\text{O}}\text{:} \\ \parallel \\ \text{:}\ddot{\text{O}}\text{:} \end{array} \right]^{2-}$ $\left[\begin{array}{c} \text{:}\ddot{\text{O}}\text{:} \\ \parallel \\ \text{:}\ddot{\text{O}}-\text{C}=\ddot{\text{O}}\text{:} \\ \parallel \\ \text{:}\ddot{\text{O}}\text{:} \end{array} \right]^{2-}$ </div> <p>Which of the following statements is true?</p> <p>A. The carbon atom is sp^2 hybridised</p> <p>B. All three carbon-oxygen bonds are equal length</p> <p>C. There are two sigma and two pi bonds in the carbonate ion</p> <p>D. There are two coordinate covalent bonds in the carbonate ion</p>	

Chemical Bond Multiple Choice Questions

Daniela Niemeyer



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there are no ... 1 Any suitable answer that refers to making space for more plants and animals as ... Answer Key Workbook 2
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