

Chemical Reactions

Atoms retain their identity.

Reactions involve only electrons and usually only outermost electrons.

Reaction rates can be increased by raising the temperature.

The energy absorbed or given off in reactions is comparatively small.

Mass is conserved. The mass of products equals the mass of starting materials.

Nuclear Reactions

Atoms usually change their identity—from one element to another.

Reactions involve mainly protons and neutrons. It does not matter what the valence electrons do.

Reaction rates are unaffected by changes in temperature.

Reactions sometimes involve enormous changes in energy.

Huge changes in energy are accompanied by measurable changes in mass ($E = mc^2$).

Chemical And Nuclear Reactions Study Guide

Yicheng Fang



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