

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

1. Li_2S Lithium Sulfide
2. CaO Calcium Oxide
3. NaF Sodium Fluoride
4. CaBr_2 Calcium Bromide
5. MgCl_2 Magnesium Chloride
6. BBr_3 Boron Bromide
7. Cs_2O Cesium Oxide
8. FrBr Francium Bromide
9. Ag_2S Silver Sulfide

10. GeF_4 Germanium Fluoride
11. Ga_2O_3 Gallium Oxide
12. EsCl_3 Einsteinium Chloride
13. Fm_2O_3 Fermium Oxide
14. Mg_3N_2 Magnesium Nitride
15. Rb_2O Rubidium Oxide
16. RaO Radium Oxide
17. SrO Strontium Oxide
18. Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

1. sodium chloride NaCl
2. magnesium fluoride MgF_2
3. silver oxide Ag_2O
4. indium bromide InBr_3
5. zinc bromide ZnBr_2
6. neodymium oxide Nd_2O_3
7. thorium sulphide ThS_2
8. actinium oxide Ac_2O_3
9. radium bromide RaBr_2
10. cesium oxide Cs_2O

11. hydrogen oxide H_2O
12. francium nitride Fr_3N
13. rubidium phosphide Rb_3P
14. potassium oxide K_2O
15. beryllium sulphide BeS
16. lithium sulphide Li_2S
17. hydrogen bromide HBr
18. strontium nitride Sr_3N_2
19. calcium oxide CaO
20. tantalum nitride Ta_3N_5

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RC Schank

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