

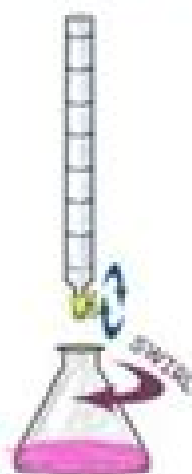
To titrate hydrochloric acid (HCl) against sodium hydroxide (NaOH) and prepare a sample of salt NaCl

- Stage 1

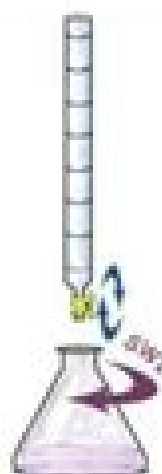


- Stage 2

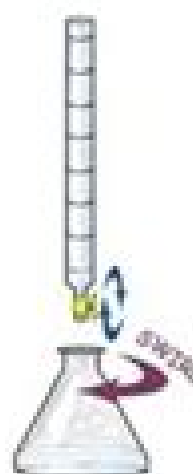
**Titration of an
Acid with a Base**
using phenolphthalein indicator



1

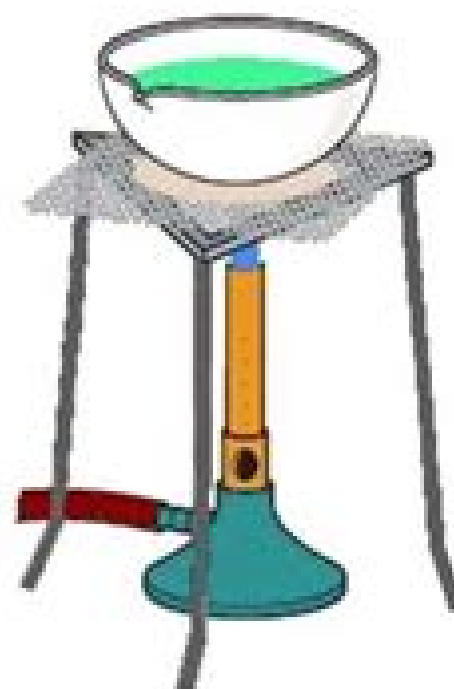


2



3

- Stage 3



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Acid-base Titrations in Nonaqueous Solvents James Sherwood Fritz, 1952 *Identification of Misconceptions Concerning Acid-base Titrations* Lockwood Ryerson Gibbs, 1995 **Acid/Base Titrations** Michael O'Brien, 1999 *PH, Buffers and Acid-base Titrations* LaMotte Chemical Products Co. Educational Products Division, 197? *Aqueous Acid-base Equilibria and Titrations* Robert De Levie, 2023 This text will give students a thorough grounding in P H and associated equilibrium material absolutely fundamental to the understanding of many aspects of chemistry This book uses the new theoretical developments that have led to more generalized approaches to equilibrium problems which are often simpler than the approximations which they replace

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