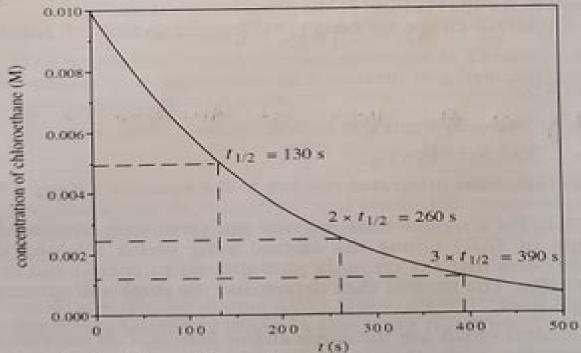
Model 2: A Simple Decomposition Reaction.

Chloroethane decomposes at 800 K:

$$CH_3CH_2Cl(g)$$
 \longrightarrow $C_2H_4(g) + HCl(g)$

The reaction is first-order with respect to chloroethane.

Figure 1. The concentration of chloroethane versus time at 800 K.



Critical Thinking Questions

4 a) Complete this table showing (CH3CH2Cl) at various times :

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