

# Balancing Chemical Equations

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STANDARD 1 LEARNING MODULE 10

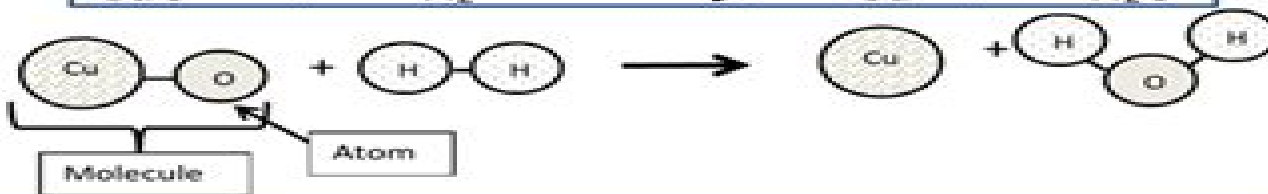
**Background Information and Why:** Chemical reactions can be described by chemical equations. Equations means equal. One each side of the equal side, also known as "yields", the number AND type of atoms are exactly the same. The Law of Conservation of Mass states that the atoms cannot be created or destroyed. The atoms can only be rearranged into different molecules/substances. Atoms are the building blocks of molecules. Therefore, in order to obey The Law of Conservation of Mass, the number of atoms for each element must be the same in the reactants (left side) as it is in the products (right side). In order to balance equations we cannot split or add to existing molecules. We can change the coefficient, adding molecules. We cannot change the subscripts, the small numbers, because they determine how many atoms are in a molecule.

## REACTION A

Chemical Equation



Atomic Representation



## REACTION B

Chemical Equation



Atomic Representation



## Questions

0.) Label one more molecule and one more atom in EACH reaction. Use the ones done above to help you. Pick another atom draw an arrow to it. Label it "atom". Then circle or "bracket" another molecule and label it "molecule".

1.) How many of each type of atom are represented in each side of reaction A? Count them up and fill out the table below.

# Atoms in Reactants	Atoms in Products
Cu- <b>ONE</b>	Cu- <b>ONE</b>
O- <b>ONE</b>	O- <b>ONE</b>
H- <b>TWO</b>	H- <b>TWO</b>

# Chem 116 Pogil Balancing Redox Equations Key

**Malika Ammam**



## **Chem 116 Pogil Balancing Redox Equations Key:**

**Balancing Chemical Equations Worksheet** Crispin Collins,2020-09-12 Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill

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Principles of Redox Reactions Malika Ammam,2018-03-05 As a teacher of physical chemistry I noticed that students even in advanced classes have difficulties in understanding the basics of redox chemistry In this Section 3 I attempted to discuss some fundamental principles related to redox processes by focusing on the species that might lose or gain electrons determination of the oxidation numbers or states of atoms in compounds and ways of balancing redox reactions To further clarify the discussed concepts numerous questions and problems with detailed answers are provided Most of these questions are formulated by students like you I believe that this Section 3 would greatly help students with levels varying from high school to advanced university classes

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