

## **REVIEW TOPICS: CHEM 141 PLACEMENT EXAM**

### **1. Chemistry: An Introduction**

- Scientific Method: Solving problems using scientific approach; observations, hypothesis, theory and laws
- Measurements: Metric system, Fundamental units, Uncertainty of measurement and Significant figures, Rounding-off numbers, Scientific notation, Derived units (eg: density), Temperature Conversions, Dimensional analysis.

### **2. Matter and Energy**

- Definition of matter
- Classification: elements and compounds; mixtures and pure substances.
- Properties: physical and chemical properties; extensive and intensive properties; physical and chemical changes
- Energy: Types of energy (kinetic energy and potential energy)

### **3. The Periodic Table**

- History: Introduction to periodic table and natural states of elements
- Symbols
- Periods and groups
- Metals, non-metals and metalloids

### **4. Atomic Structure**

- Sub-atomic particles
- Atomic mass
- Isotopes
- Law of conservation of mass; Law of multiple proportions; Law of constant composition
- Dalton's atomic theory
- Formulas of compounds (define and interpret formulas)

### **5. Nomenclature**

- Differentiate ionic and covalent bonding
- Ions (anions and cations) and naming ions
- Compounds and naming compounds (naming type I and type II ionic compounds)
- Naming covalent molecular species
- Polyatomic ions and naming compounds containing polyatomic ions.
- Naming acids and bases
- Writing formulas and names

### **6. Types of Reactions**

- Evidence for a chemical reaction and chemical equation
- Writing and balancing chemical reactions
- Dissolution of compounds
- Acid-Base reactions
- Classification/recognizing chemical reactions.

### **7. Chemical Composition and Chemical Quantities**

- Counting by weighing: atomic mass (counting atoms by weighing)
- The mole concept: molar mass; percent composition
- Formulas of compounds (formula masses, percent composition)
- Understanding empirical and molecular formulas
- Information given by a chemical equation;
- Mole-mole calculations and concept of limiting reagents

### **8. Electronic Configuration**

- Bohr Theory of Structure of an atom.
- Modern concept of atomic structure: shells, subshells and orbitals, shapes of orbitals
- Electronic Configurations

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