

Chemistry 1212
Study Guide for Final Exam

1) Chapter 11 Liquids, Solids, and Intermolecular forces

a) Review Intermolecular Forces

$$1) E = \frac{1}{4\pi\epsilon_0} \frac{q_1 q_2}{r}$$

b) Dispersion forces, Dipole-Dipole, Hydrogen Bonding, Ion-Dipole, Surface Tension, Viscosity, Capillary Action.

c) Vaporization and Vapor Pressure, Dynamic Equilibrium

1) Clausius-Clapeyron Equations

$$a) \ln P_{\text{vap}} = \frac{-\Delta H_{\text{vap}}}{R} \left(\frac{1}{T} \right) + \ln \beta$$

$$b) \ln \frac{P_2}{P_1} = \frac{-\Delta H_{\text{vap}}}{R} \left(\frac{1}{T_2} - \frac{1}{T_1} \right) \quad \text{Two-point form}$$

d) Critical Point

e) Fusion and Sublimation

f) Phase Diagrams

g) Intermolecular Forces and Water

h) Band Theory (Crystalline Structures)

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

2) Chapter 12 Solutions

a) Solutions

1) Types

2) Solubility of substances

3) Entropy as it relates to solubility

4) Effects of intermolecular forces on solubility

b) Energetics of solution formation

$$a) \Delta H_{\text{soln}} = \underbrace{\Delta H_{\text{solute}}}_{\text{endothermic (+)}} + \underbrace{\Delta H_{\text{solvent}}}_{\text{endothermic (+)}} + \underbrace{\Delta H_{\text{mix}}}_{\text{exothermic (-)}}$$

1) Heats of hydration (See the relationships on pages 446 & 447)

2) Solution Equilibrium

3) Solution Concentration Methods

4) Gas as a solute (Henry's Law)

5 Colligative Properties

a) Freezing point depression and Boiling point elevation

I recommend that you review your quizzes and try a couple of problems under each section of the Exercises.

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Wolfgang Guggemos



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