He Key Re Currents	Deterance Sheet	Cattornia Standards Test
No.   Common symbol   Common	Name	15

A solution of AlCl<sub>3</sub> and K<sub>2</sub>CO<sub>3</sub> results in the formation of Al<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub> and KCl. The balanced chemical equation is as follows:

What is the theoretical yield of Al<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub> (in grams), that is produced when 47.4 ½ of AlCl<sub>3</sub> react with 46.7 g of K<sub>2</sub>CO<sub>3</sub>?

# Select one:

- The theoretical yield of Al2(CO3)3 is 26.3 grams.
- The theoretical yield of Al2(CO3)3 is 41.5 grams.
- The theoretical yield of Al2(CO3)3 is 13.1 grams.
- The theoretical yield of Al2(CO3)3 is 13.3 grams.
- The theoretical yield of Al2(CO3)3 is 78.840 grams.

# **California Standards Test Chemistry Answer Key**

**Greg Branish** 

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