

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

At your notebook, read about suspensions and colloids.

For each statement below, write true or false.

- _____ 1. A solution is a mixture consisting particles that settle out of the mixture if left undisturbed.
- _____ 2. The most abundant substance in a colloid is the dispersion medium.
- _____ 3. A colloid can be separated by filtration.
- _____ 4. A solid emulsion consists of a liquid dispersed in a solid.
- _____ 5. Whipped cream is an example of a foam.
- _____ 6. In an aerosol, the dispersing medium is a liquid.
- _____ 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- _____ 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- _____ 9. Adding an electrolyte into a colloid stabilizes the colloid.
- _____ 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture where particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
01. Less than 1 nm in diameter			
02. Between 1 nm and 1000 nm in diameter			
03. More than 1000 nm in diameter			
04. Settle out if undisturbed			
05. Pass through standard filter paper			
06. Largest visible particles			
07. Scatter light			

Chapter 14 Study Guide For Content Mastery Answer Key

A Gutmann



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