

CHAPTER 15 STUDY GUIDE

Energy and Chemical Change

Section 15.1 Energy

In your textbook, read about the nature of energy.

In the space at the left, write *true* if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

- _____ 1. *Energy* is the ability to do work or produce heat.
- _____ 2. The law of conservation of energy states that energy *can be* created and destroyed.
- _____ 3. Chemical potential energy is energy stored in a substance because of its *composition*.
- _____ 4. *Heat* is a form of energy that flows from a warmer object to a cooler object.
- _____ 5. A calorie is the amount of energy required to raise the temperature of *one gram* of pure water by one degree Celsius.
- _____ 6. A *calorie* is the SI unit of heat and energy.
- _____ 7. The *specific heat* of a substance is the amount of heat required to raise the temperature of one gram of that substance by one degree Celsius.
- _____ 8. *Kinetic energy* is energy of motion.
- _____ 9. Chemicals participating in a chemical reaction contain only *potential energy*.
- _____ 10. One nutritional Calorie is equal to *100 calories*.
- _____ 11. One calorie equals *4.184 joules*.
- _____ 12. When a fuel is burned, some of its *chemical potential energy* is lost as heat.
- _____ 13. To convert kilojoules to joules, *divide* the number of kilojoules by 1000 joules/1 kilojoule.

Answer the following question. Show all your work.

14. If the temperature of a 500.0-g sample of liquid water is raised 2.00°C, how much heat is absorbed by the water? The specific heat of liquid water is 4.184 J/(g·°C).

Chapter 15 Study Guide Energy And Chemical Change Answers

DP Hallahan



Chapter 15 Study Guide Energy And Chemical Change Answers:

Whispering the Strategies of Language: An Emotional Journey through **Chapter 15 Study Guide Energy And Chemical Change Answers**

In a digitally-driven earth wherever monitors reign great and quick conversation drowns out the subtleties of language, the profound strategies and mental subtleties concealed within words frequently move unheard. Yet, located within the pages of **Chapter 15 Study Guide Energy And Chemical Change Answers** a fascinating literary prize blinking with fresh feelings, lies an extraordinary quest waiting to be undertaken. Penned by a talented wordsmith, this wonderful opus attracts viewers on an introspective trip, softly unraveling the veiled truths and profound affect resonating within the material of each and every word. Within the emotional depths with this moving review, we will embark upon a honest exploration of the book is core subjects, dissect its captivating writing design, and succumb to the powerful resonance it evokes strong within the recesses of readers hearts.

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