Practice

- 1) TiO₂ + C + 2Cl₂ → TiCl₄ + CO₂
 - -If you begin with 1.25 mol TiO₂, what mass of Cl₂ gas is needed?



<u>Chapter 12 Study Guide For Content Mastery</u> <u>Stoichiometry Answers</u>

Vincent Verret

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