

Combustion Phenomena

- Activation energy: The amount of energy needed to get a chemical reaction started.
- Fuel molecules and oxygen atoms merely next to each other don't just spontaneously ignite.
 - Spark excites fuel and oxygen so they will chemically combine, starts burning.
 - Fuel and oxygen molecule must be properly lined up.
 - Fuel molecule burns from one end like a zipper.
 - Heat liberated starts next fuel/oxygen pair burning.
 - Chain reaction. One reaction passes energy to the next.
 - Heat liberated in the chemical reaction causes gases to expand, this is what forces the piston down.

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Yan Bai



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