

7. What is a catalyst? Explain why the presence of a catalyst increases the rate of a reaction.

8. A biological catalyst used in living things is called an _____.

9. An inhibitor is the opposite of a catalyst; it is a substance that slows down or prevents a chemical reaction from occurring. Describe three uses of inhibitors.

10. Using ideas from collision theory, explain the following statement.

In our experiment, we prepared 2 different batches. One batch contained 500 mL of 10% hydrochloric acid and the other contained 500 mL of 10% dilute hydrochloric acid. We placed 1g of Alka-Seltzer tablet into each batch and then timed until effervescent stopped. The Alka-Seltzer tablet fully reacted in 1.5 seconds in the first solution and 3.6 seconds in the second.

11. Using ideas from collision theory, explain the following statement.

When Alka-Seltzer is an effervescent tablet it contains citric acid. When the tablet is dissolved in water, carbon dioxide bubbles come from the air (20%) and hydroxide gas (80%) from dissociation in the hydrochloric acid ($Na + Cl \rightarrow NaCl$). The reaction is governed by Le Chatelier's principle and an equilibrium shift in which carbonic acid dissociates to form carbon dioxide.

Collision Theory 161 Answer Key

DP Hallahan

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