

4. a. A sample of a  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O} / \text{K}_2\text{C}_2\text{O}_4 \cdot \text{H}_2\text{O}$  solid salt mixture is dissolved in ~150 mL of deionized water previously adjusted to a pH that is basic. The precipitate, after having been filtered, was air-dried and weighed. Data for Trial 1 were obtained as shown. Complete the following table. (See *Report Sheet*.) Record calculated values with the correct number of significant figures.

**A. Precipitation of  $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$  from the Salt Mixture**

**Calculation Zone**

3. Mass of salt mixture (g)	<u>0.879</u>
4. Mass of filter paper (g)	<u>1.896</u>
5. Mass of filter paper and $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u>2.180</u>
6. Mass of air-dried $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ (g)	<u>          </u>

*Data Analysis, 1.*

**B. Determination of Limiting Reactant**

1. Limiting reactant in salt mixture	<u><math>\text{CaCl}_2 \cdot 2\text{H}_2\text{O}</math></u>
2. Excess reactant in salt mixture	<u>          </u>

*Data Analysis, 3.*

**Data Analysis**

1. Moles of $\text{CaC}_2\text{O}_4 \cdot \text{H}_2\text{O}$ precipitated (mol) <i>Show calculation.</i>	<u>          </u>
2. Moles of limiting reactant in salt mixture (mol) <i>See equation 8.1.</i>	<u>          </u>
3. Mass of limiting reactant in salt mixture (g) <i>Show calculation.</i>	<u>          </u>
4. Mass of excess reactant in salt mixture (g) <i>Equals mass of salt mixture minus mass of limiting reactant. Show calculation.</i>	<u>          </u>
5. Percent limiting reactant in salt mixture (%) <i>Show calculation.</i>	<u>          </u>
6. Percent excess reactant in salt mixture (%)	<u>          </u>

*Data Analysis, 4.*

*Data Analysis, 5.*

4. b. For Trials 2 and 3, the percent  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the mixture was 39.7% and 28.4% respectively.

a. What is the average percent of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the sample? **Data Analysis, B.**

b. What are the standard deviation and the relative standard deviation (%RSD) for the percent of  $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$  in the sample? **Data Analysis, C and D.**

## Exp 8 Limiting Reactant Prelab Answers

**John R. Holum, Ruth C. Denison, Sandra  
L. Olmsted**



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