

The components in a heterogeneous mixture can be seen easily and can be separate them by physical means.

**For Example:** If we mix sand in water the sand settles down in water after some time and we can separate it by filtration.

#### DIFFERENCES - HOMOGENEOUS & HETEROGENEOUS MIXTURES

Sr.	Homogenous Mixtures	Heterogeneous Mixtures
1.	They have a uniform composition throughout	They have a non-uniform composition
2.	components of the mixture cannot separated through physical processes	Components can be separate through physical processes.
3.	Components cannot be seen through naked eyes.	Components can easily be seen through naked eyes.
4.	The mixture is in single phase throughout.	The substances can be of two different phases and we may see separate layers of the substances
5.	Example: A mixture of water and milk	Example: A mixture of oil in water

#### 2.4| DIFFERENCES- MIXTURES AND COMPOUNDS

Sr.	Mixtures	Compounds
1.	Combination of two or more substances in any ratio.	Combination of two or more substances in fixed ratio.
2.	Components reflects the properties of mixture.	Compound has entire different properties from its constituents.
3.	Its components Can be separated by physical means.	Its components Cannot be separated by physical means.

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