



GLOBAL REASONING TEST PRACTICE TEST ANSWER SHEET



Practise Test Start Time: _____ Practise Test End Time: _____

TEST INSTRUCTIONS

- You must remain seated throughout the test.
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Practise Test



ANSWER SHEET

Question 1	Question 2	Question 3
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E
☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E	☐ A ☐ B ☐ C ☐ D ☐ E

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Christian G. Meyer



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web the mole is an amount unit similar to familiar units like pair dozen gross etc it provides a specific measure of the number of atoms or molecules in a sample of matter one latin connotation for the word mole is large mass or bulk which is consistent with its use as the name for this unit

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web libretexts the mole concept is central to any chemistry calculation based on experimental results the mole is how we relate the unbelievably small atoms and molecules that make something up to the measurable properties such as mass which

we may observe in a laboratory setting

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web the mole was made the seventh si base unit in 1971 by the 14th cgpm 2019 redefinition of si base units before the 2019 redefinition of the si base units the mole was defined as the amount of substance of a system that contains as many elementary entities as there are atoms in 12 grams of carbon 12 the most common isotope of carbon