

AP Biology Chapter Questions - Campbell 7th Edition

Modified from Campbell and Reece Instructors Guide

Chapter 1: Exploring Life

Exploring Life on Its Many Levels

1. Briefly describe the unifying concepts that characterize living organisms.
2. Diagram the hierarchy of structural levels in biological organization.
3. Explain how the properties of life emerge from complex organization.
4. Distinguish between prokaryotic and eukaryotic cells.
5. Explain the importance of regulatory mechanisms in living things. Distinguish between positive and negative feedback.

Evolution, Unity, and Diversity

6. Distinguish among the three domains of life. List and distinguish among the three kingdoms of multicellular, eukaryotic life.
7. Explain why diagrams of evolutionary relationships have a treelike form.

The Process of Science

8. Distinguish between quantitative and qualitative data.
9. Distinguish between inductive and deductive reasoning.
10. Distinguish between Reductionism and Systems Biology.
11. Explain why hypotheses must be testable and falsifiable but are not provable.
12. Describe what is meant by a controlled experiment.
13. Distinguish between the everyday meaning of the term theory and its meaning to scientists.
14. Explain how science is influenced by social and cultural factors.

Chapter 2: The Chemical Context of Life

Elements and Compounds

1. Distinguish between an element and a compound.
2. Identify the four elements that make up 96% of living matter.
3. Define the term trace element and give an example.

Atoms and Molecules

4. Draw and label a simplified model of an atom. Explain how this model simplifies our understanding of atomic structure.
5. Distinguish between each of the following pairs of terms:
 - a. neutron and proton
 - b. atomic number and mass number
 - c. atomic weight and mass number
6. Explain how two isotopes of an element are similar. Explain how they are different.
7. Describe two biological applications that use radioactive isotopes.
8. Distinguish among nonpolar covalent, polar covalent and ionic bonds.
9. Explain why strong covalent bonds and weak bonds are both essential in living organisms.
10. Distinguish between hydrogen bonds and van der Waals interactions.
11. Give an example that illustrates how a molecule's shape can determine its biological function.
12. Explain what is meant by a chemical equilibrium.

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RJ Shavelson



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