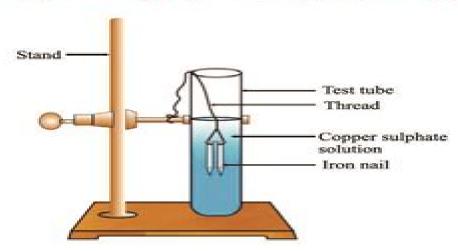
- * Above reaction is used in black & white photography.
 - Endothermic Reactions: The reactions which require energy in the form of heat, light or electricity to break reactants are called endothermic reactions.
- III. DISPLACEMENT REACTION: The chemical reaction in which more reactive element displaces less reactive element from its salt solution.

Fe (s) +
$$CuSO_4$$
 (aq) \rightarrow $FeSO_4$ (aq) + Cu (s)



The iron nail becomes brownish in colour by deposition of Cu and blue colour of CuSO₄ changes to dirty green colour due to formation of FeSO₄.

Zn is more reactive than copper.

IV. DOUBLE DISPLACEMENT REACTION: A reaction in which new compounds are formed by mutual exchange of ions between two compounds.

$$Na_2SO_4$$
 (aq) + $BaCl_2$ (aq) \rightarrow $BaSO_4$ (s) + $2NaCl$ (aq) (Sodium sulphate) (Barium (Barium (Sodium chloride) sulphate)



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Victor M. Corman

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