



Science Lab Manual



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Class IX

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A. K. Raj



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SPOTTING 1 Pollination in flowers 2 Pollen germination 3 Slides of mammal tissues 4 Meiosis cell division 5 T S of Blastula 6 Mendel s inheritance laws 7 Pedigree chart 8 Controlled pollination 9 Common disease causing organisms 10 Symbolic Association in root nodules 11 Homologous and analogous organs PROJECTS 1 To study the different means of pollination 2 To study infectious diseases of humans 3 To study birth rate and death rate In your village or town 4 To study genetic disorders 5 To study malaria causes and disorders 6 To study causes symptoms and diagnosis of cancer 7 To study causes symptoms and diagnosis of AIDS 8 To study the applications and importance of Biotechnology 9 At the time of COVID 19

what kind of problems did your acquaintance have to face when he got covid 10 To study DRUG ABUSE 11 Name the medicinal drugs which are banned all over the world 12 Describe the response of biotic factors to abiotic factors in the environment 13 Biodiversity and Conservation Practices in Indian Culture 14 What has been the effect on pollution control after covid 15 To study Management of sewage and waste materials 16 To study the role of micro organisms in human welfare 17 To study about harmful micro organisms VIVA VOCE FOR PRACTICE **Laboratory Manual for Science** □ 6 A. K. Raj, Laboratory Manual for Science is a series of five books for classes 6 to 10 These are complimentary to the Science textbooks of the respective classes The manuals cover a wide range of age appropriate experiments that give hands on experience to the students The experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and techniques used in scientific investigations Our manuals aim not only to help students better comprehend the scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds **Space Science, Grade 6-8**, 2011-06-08 Laboratory Manual for Science □ 8 A. K. Raj, Laboratory Manual for Science is a series of five books for classes 6 to 10 These are complimentary to the Science textbooks of the respective classes The manuals cover a wide range of age appropriate experiments that give hands on experience to the students The experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and techniques used in scientific investigations Our manuals aim not only to help students better comprehend the scientific concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds *Sci Explorer Integrated Sci Lab Man 2000c* David V. Frank, Prentice Hall Direct Education Staff, Prentice-Hall, Inc, 1999-03-01 *McDougal Littell Science* McDougal Littell, 2004-05 **McDougal Littell Science** McDougal-Littell Publishing Staff, 2004-05-14 *Quality Science Labs Grade 6 Lab Manual* Elva Burlingham, Ken Schaefer, Marlene Kent, 2014-04-15 This manual was written to meet Texas Essential Knowledge and Skills TEKS standards and to accompany a lab kit which includes supplies and equipment for each lab as well as a student journal and a teacher answer guide Lab experiments MATTER AND ENERGY 1 Elements Metals Metalloids and Nonmetals 2 Density and the Case of the Lost Gold Bar 3 Properties of Rock Forming Minerals 4 Fast Rusting and Chemical Reactions in a Baggie FORCE MOTION AND ENERGY 5 Energy Transformations 6 Roadblocks and Energies 7 Pulleys 8 Amazing Molecules in Motion EARTH AND SPACE AND ENERGY IN THE EARTH SYSTEM 9 Layers of the Earth 10 The Rock Cycle 11 Plate Tectonics 12 Finding an Earthquake s Epicenter 13 The Sun and Weather Angle of the Sun 14 Visible and Invisible Light From the Sun The EMS 15 Topography 16 Planetary Orbits 17 Gravity 18 Space Travel ORGANISMS AND ENVIRONMENTS 19 Cell Modeling Prokaryotic and Eukaryotic Cells 20 Classifications Domains and Kingdoms 21 Biotic and Abiotic Factors in a Habitat 22 Ecosystem Explorations How is an Ecosystem Organized **Practical/Laboratory Manual Chemistry Class XI based on NCERT guidelines by Dr. S. C. Rastogi & Er. Meera Goyal** Dr. S. C. Rastogi , Er. Meera Goyal, 2020-06-23 An Excellent Book in Accordance with the latest

syllabus for Class 11 Prescribed by CBSE NCERT and Adopted by Various State Education Boards
A Basic Laboratory Techniques
 1 To cut a glass tube or glass rod
 2 To bend the glass rod at an angle
 3 To draw a glass jet from a glass tube
 4 To bore a cork and fit a glass tube into it
B Characterisation and Purification of Chemical Substances
 1 To determine the melting point of the given unknown organic compound and its identification simple laboratory technique
 2 To determine the boiling point of a given liquid when available in small quantity simple laboratory method
 3 To prepare crystals of pure potash alum $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$ from the given impure sample
 4 To prepare the pure crystals of copper sulphate from the given crude sample
 5 To prepare pure crystals of benzoic acid from a given impure sample
C Measurement of pH Values
 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper
 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid CH_3COOH of same concentration
 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper
 4 To study the pH change by common ion CH_3COO^- ion in case of weak acid CH_3COOH
 5 To determine the change in pH value of weak base NH_4OH in presence of a common ion NH_4^+
D Chemical Equilibrium
 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions
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E Quantitative Analysis
 1 To prepare M 10 oxalic acid solution by direct weighing method
 2 To prepare M 10 solution of sodium carbonate by direct weighing method
 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid
 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution
F Qualitative Analysis
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 2 Analysis of Cations
G Detection of Elements in Organic Compounds
 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test
 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test
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 1 To check the bacterial contamination in drinking water by testing sulphide ions
B Methods of Water Purification
 1 To purify water from suspended impurities by using sedimentation
 2 To purify water by boiling
 3 To purify water by distillation method
 4 To purify water by reverse osmosis technique
 5 To purify water by GAC method
 6 To purify water by bleach treatment
 7 To purify water by oxidising agent
 8 To purify water by ozone treatment method
C Water Analysis
 1 To test the hardness of different water samples
D Foaming Capacity of Various Soaps
 1 To compare the foaming capacity of different washing soaps
 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap
E Tea Analysis
 1 To study the acidity of different samples of tea leaves tea by using pH paper
F Analysis of Fruits and Vegetable Juices
 1 To analyse the fruit and vegetable juices for the constituent present in them
G Rate of Evaporation
 1 To study the rate of evaporation of different liquids
H Effect of Acids and Bases on Tensile Strength of Fibres
 1 To compare the tensile strength of natural fibres and synthetic fibres
 2 To study

the effect of acids and bases on tensile strength of different fibres Log Antilog Table **McDougal Littell Science**
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 classes The manuals cover a wide range of age appropriate experiments that give hands on experience to the students The
 experiments help students verify scientific truths and principles and at the same time expose them to the basic tools and
 techniques used in scientific investigations Our manuals aim not only to help students better comprehend the scientific
 concepts taught in their textbooks but also to ignite a scientific quest in their young inquisitive minds Science, Grade 6
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