

Class 10th Chemistry Practical Copy

Roll No. _____ Page No. _____

A solution can be identified only by considering the properties and characteristics of it. In other words, the combination of most solutions are an empirical source of it which have been associated with some specific signs of it. For example, the pH value can give it the pH.

Acid solution: $[H^+] > 10^{-7} M$, pH < 7.00

Basic solution: $[OH^-] > 10^{-7} M$, pH > 7.00

Neutral solution: $[H^+] = [OH^-] = 10^{-7} M$, pH = 7.00

- A solution having a pH of 7 will be a neutral solution. Solutions with pH values above 7 are basic or alkaline and solutions with pH values below 7 are acids.
- pH of a solution can be determined using a pH meter or using an indicator with universal indicator.

UNIVERSAL INDICATOR

It is a mixture of several indicators having different pH ranges and is used to determine the approximate value of pH of a solution.

The colour that indicates the pH of a solution after adding a universal indicator is:

pH range	Indicator	Colour
0-2	Strong acid	Red
3-6	Weak acid	Orange/Yellow
7	Neutral	Green
8-11	Basic	Blue
12-14	Strong base	Violet

Source: Wikipedia

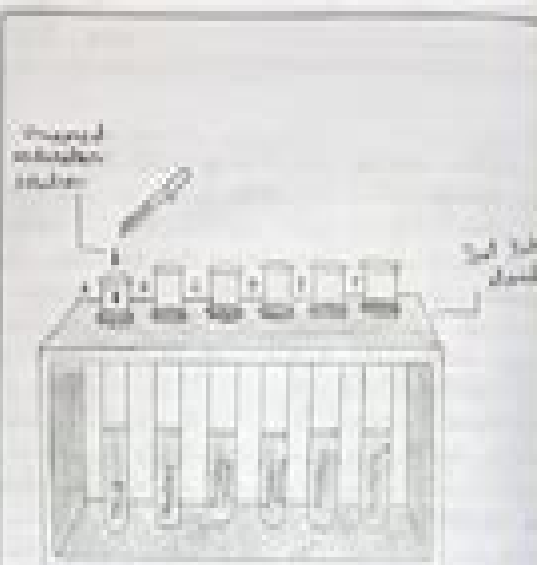
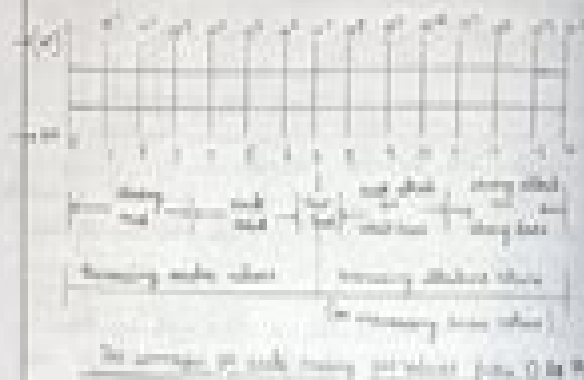


Fig. 1.1



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Class 10 Chemistry Practical Lab Manual Cbse

CH Cherryholmes



Class 10 Chemistry Practical Lab Manual Cbse:

This book delves into Class 10 Chemistry Practical Lab Manual Cbse. Class 10 Chemistry Practical Lab Manual Cbse is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Class 10 Chemistry Practical Lab Manual Cbse, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Class 10 Chemistry Practical Lab Manual Cbse
- Chapter 2: Essential Elements of Class 10 Chemistry Practical Lab Manual Cbse
- Chapter 3: Class 10 Chemistry Practical Lab Manual Cbse in Everyday Life
- Chapter 4: Class 10 Chemistry Practical Lab Manual Cbse in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Class 10 Chemistry Practical Lab Manual Cbse. The first chapter will explore what Class 10 Chemistry Practical Lab Manual Cbse is, why Class 10 Chemistry Practical Lab Manual Cbse is vital, and how to effectively learn about Class 10 Chemistry Practical Lab Manual Cbse.
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4. In chapter 3, the author will examine the practical applications of Class 10 Chemistry Practical Lab Manual Cbse in daily life. The third chapter will showcase real-world examples of how Class 10 Chemistry Practical Lab Manual Cbse can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Class 10 Chemistry Practical Lab Manual Cbse in specific contexts. This chapter will explore how Class 10 Chemistry Practical Lab Manual Cbse is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Class 10 Chemistry Practical Lab Manual Cbse. The final chapter will summarize the key points that have been discussed throughout the book.
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