

The Periodic Table of the Elements

The Periodic Table of the Elements

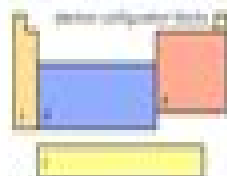
The periodic table displays elements organized by atomic number (1 to 118). A callout box for Iron (Fe) provides the following data:

- atomic mass: 55.845
- atomic number: 26
- 1st ionization energy: 762.5 kJ/mol
- electronegativity: 1.83
- chemical symbol: Fe
- name: Iron
- electron configuration: $[Ar] 3d^6 4s^2$
- oxidation states: +2, +3, +6

Legend for element categories:

- alkali metals (orange)
- alkaline earth metals (yellow)
- other metals (light green)
- transition metals (green)
- lanthanoids (light blue)
- actinoids (dark blue)
- metalloids (light purple)
- nonmetals (purple)
- halogens (pink)
- noble gases (light blue)
- unknown elements (grey)
- radioactive elements (yellow with black border)

1 H Hydrogen																	18 He Helium														
3 Li Lithium	4 Be Beryllium											11 Na Sodium	12 Mg Magnesium							13 Al Aluminum	14 Si Silicon	15 P Phosphorus	16 S Sulfur	17 Cl Chlorine	18 Ar Argon						
19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton														
37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon														
55 Cs Cesium	56 Ba Barium	57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium	71 Lu Lutetium	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon
87 Fr Francium	88 Ra Radium	89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium	103 Lr Lawrencium	104 Rf Rutherfordium	105 Db Dubnium	106 Sg Seaborgium	107 Bh Bohrium	108 Hs Hassium	109 Mt Meitnerium	110 Ds Darmstadtium	111 Rg Roentgenium	112 Cn Copernicium	113 Nh Nihonium	114 Fl Flerovium	115 Mc Moscovium	116 Lv Livermorium	117 Ts Tennessine	118 Og Oganesson



Notes:

- 1. Not all elements (113-118) have an official name designation (the suffix).
- 2. Elements in the red box are synthetic.
- 3. All elements are subject to further scientific study.

57 La Lanthanum	58 Ce Cerium	59 Pr Praseodymium	60 Nd Neodymium	61 Pm Promethium	62 Sm Samarium	63 Eu Europium	64 Gd Gadolinium	65 Tb Terbium	66 Dy Dysprosium	67 Ho Holmium	68 Er Erbium	69 Tm Thulium	70 Yb Ytterbium
89 Ac Actinium	90 Th Thorium	91 Pa Protactinium	92 U Uranium	93 Np Neptunium	94 Pu Plutonium	95 Am Americium	96 Cm Curium	97 Bk Berkelium	98 Cf Californium	99 Es Einsteinium	100 Fm Fermium	101 Md Mendelevium	102 No Nobelium

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TD Snyder



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Dimstri Chemistry of Elements P. Subarat, 2015-05-18 New Researched Findings In Field of Chemical Sciences

Chemistry of the Elements Norman Neill Greenwood, 1993 **Chemistry of the Elements** ,2005 **30-second**

Elements Eric Scerri, 2014 The chemical elements are the building blocks of life but could you discuss the periodic table over the dinner table Which elements put the blue into blu ray and the lime into limelight And do you know enough about antimony arsenic and aluminium to illuminate the bar with your elemental knowledge 30 Second Elements presents you with the foundations of chemical knowledge distilling the 50 most significant chemical elements into half a minute individual entries using nothing more than two pages 300 words and one picture Divided into seven chapters it includes the atomic details of the other 68 elements and the relationships of all 118 as well as biographies of the chemists who transformed scientific knowledge and unlocked the mysteries of life itself Illustrated with explosive graphics here is the quickest way to know your arsenic from your europium **30-Second Elements** Eric Scerri, 2013-08-01 When was radium discovered Who

are Dmitri Mendeleev and Glenn T Seaborg Who discovered uranium s radioactivity Which element is useful for dating the age of Earth And why doesn t gold have a scientific name 30 Second Elements presents you with the very foundations of chemical knowledge explaining concisely the 50 most significant chemical elements This book uses helpful glossaries and tables to fast track your knowledge of the other 68 elements and the relationships between all of them **The Elements** Philip Ball, 2021-09-27 The classical elements The antique metals Alchemical elements The new metals Chemistry golden age Electrical discoveries The radiant age The nuclear age [A Tale of Seven Elements](#) Eric Scerri, 2013-07-18 In A Tale of Seven Elements Eric Scerri presents the fascinating history of those seven elements discovered to be mysteriously missing from the periodic table in 1913 *Chemistry of the Elements* Norman Neill Greenwood, 1997-10 **A Tale of Seven**

Elements Eric R. Scerri, 2025 In 1913 English physicist Henry Moseley transformed our understanding of the elements by organizing them by atomic number from hydrogen to uranium However it soon became obvious that researchers had yet to discover seven completely unknown elements In this updated and extended edition of his acclaimed book Eric Scerri presents the intriguing stories of these elements protactinium hafnium rhenium technetium francium astatine and promethium It follows their discoveries through the two world wars starting with protactinium in 1917 and ending with promethium in 1945 Scerri details the research leading to each discovery the pivotal experiments the chemists involved the properties of each element and their applications For example hafnium alloys named after Copenhagen hafnia have some of the highest boiling points and are used in rocket thrusters like those on the Apollo Lunar Modules Scerri then examines the integration of twenty five new elements into the periodic table over the past fifty years and concludes with a new chapter on rare Earth elements and the histories of their discoveries The Second Edition of A Tale of Seven Elements vividly portrays the journey of chemical research from missteps and disputes to the thrill of discovery illuminating the researchers behind the elements as

much as their breakthroughs Provided by publisher **What Is A Chemical Element?** Eric Scerri,Elena Ghibaudi,2020-05-11 The concept of a chemical element is foundational within the field of chemistry but there is wide disagreement over its definition Even the International Union for Pure and Applied Chemistry IUPAC claims two distinct definitions a species of atoms versus one which identifies chemical elements with the simple substances bearing their names The double definition of elements proposed by the International Union for Pure and Applied Chemistry contrasts an abstract meaning and an operational one Nevertheless the philosophical aspects of this notion are not fully captured by the IUPAC definitions despite the fact that they were crucial for the construction of the Periodic Table Although rich scientific literature on the element and the periodic table exists as well as a recent growth in the philosophy of chemistry scholars are still searching for a definitive answer to this important question What is an element Eric Scerri and Elena Ghibaudi have teamed up to assemble a group of scholars to provide readers an overview of the current state of the debate on chemical elements from epistemological historical and educational perspectives What Is A Chemical Element fills a gap for the benefit of the whole chemistry community experimental researchers philosophers chemistry educators and anyone looking to learn more about the elements of the periodic table **The Lost Elements** Marco Fontani,Mariagrazia Costa,Mary Virginia Orna,2014-10-01 The Periodic Table of Elements hasn't always looked like it does now a well organized chart arranged by atomic number In the mid nineteenth century chemists were of the belief that the elements should be sorted by atomic weight However the weights of many elements were calculated incorrectly and over time it became clear that not only did the elements need rearranging but that the periodic table contained many gaps and omissions there were elements yet to be discovered and the allure of finding one had scientists rushing to fill in the blanks Supposed discoveries flooded laboratories and the debate over what did and did not belong on the periodic table reached a fever pitch With the discovery of radioactivity the discourse only intensified Throughout its formation the Periodic Table of Elements has seen false entries good faith errors retractions and dead ends In fact there have been more falsely proclaimed elemental discoveries throughout history than there are elements on the table as we know it today The Lost Elements The Periodic Table's Shadow Side collects the most notable of these instances stretching from the nineteenth century to the present The book tells the story of how scientists have come to understand elements by discussing the failed theories and false discoveries that shaped the path of scientific progress We learn of early chemists stubborn refusal to disregard alchemy as a legitimate practice and of one German's supposed discovery of an elemental metal that breathed As elements began to be created artificially in the twentieth century we watch the discovery climate shift to favor the physicists rather than the chemists Along the way Fontani Costa and Orna introduce us to the key figures in the development of today's periodic table including Lavoisier and Mendeleev Featuring a preface from Nobel Laureate Roald Hoffmann The Lost Elements is an expansive history of the wrong side of chemical discovery and reveals how these errors and gaffes have helped shape the table as much as any other form of

scientific progress *The Discovery of the Periodic Table of the Chemical Elements* Torsten Schmiernund, 2022 150 years ago in 1869 D I Mendeleev and L Meyer independently published their ideas on the arrangement of the chemical elements in a periodic system The United Nations and UNESCO therefore declared 2019 the International Year of the Periodic Table The question arises what is so special about this simple table Join the author on a short journey to the history of the periodic table Learn about its predecessors and look at how the periodic table of elements has evolved over the years Discover the periodic properties of the elements Learn what makes the periodic table so interesting and timeless and see what other ideas there are and have been for representing it The content Atomic theories atomic structure and atomic masses Early attempts to systematize the elements Mendeleev s periodic table and periodicities of the elements Alternative periodic systems The target groups Students and lecturers of chemistry especially history of chemistry Teachers and students as well as those interested in the history of chemistry The author Torsten Schmiernund has been working as a chemical technician in the chemical industry for many years This book is a translation of an original German edition The translation was done with the help of artificial intelligence machine translation by the service DeepL com A subsequent human revision was done primarily in terms of content so that the book will read stylistically differently from a conventional translation **Discovery of the Elements** Mary Elvira Weeks, 1943 The Elements Sir William Augustus Tilden, 1910 Elements Kennon O'Mara, 2018-12-15 Elements create everything around us from the air we breathe to our very bodies Some elements are abundant such as hydrogen and carbon while others are rare or only created in laboratories How elements join to create molecules and compounds and the matter around us is an amazing discovery that scientists are still investigating Readers will appreciate the accessible look at elements in this intriguing book which includes comprehensible fact boxes helpful diagrams and essential elementary chemistry facts **The Aqueous Chemistry of the Elements** George K. Schweitzer, Lester L. Pesterfield, 2010-01-14 This volume is a comprehensive treatment of the aqueous solution chemistry of all the elements An E pH diagram for each element sets the context for the chemistry of that element Elements, Compounds and Mixtures Brian J. Knapp, 1998 Elements compounds and mixtures Chemlab The Elements , 1989 Reference book containing the most important facts about all the chemical elements arranged in alphabetical order *Exploring the Chemistry of the Elements* John R. Moss, 1995 *The Periodic Table of Elements Compendium* Helen Eccles, 1994 This book provides descriptions of 103 elements their properties occurrences and uses plus a summary of periodic table patterns

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