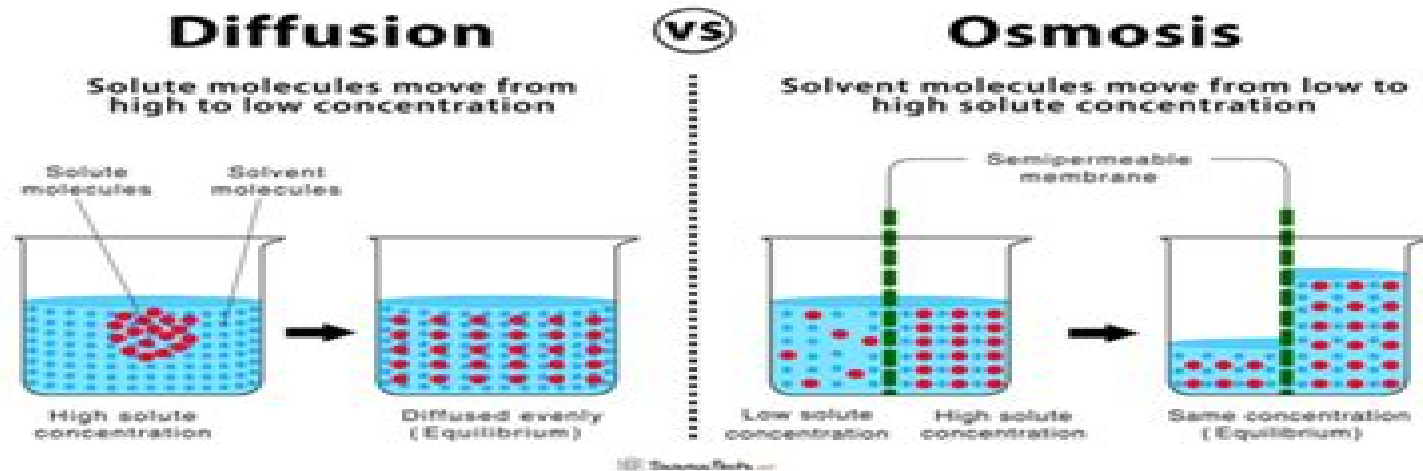


Diffusion and Osmosis

Background:

Diffusion is the passive movement of particles of a solute from an area of high concentration to an area of low concentration.

Osmosis is the passive movement of a solvent, such as water, through a semipermeable membrane from a region of high solvent concentration to an area of low solvent concentration.



Solute: a substance that has been dissolved in a given solution.

Solvent: a substance, usually a liquid, that is capable of dissolving or dispersing one or more other substances.

Ex: To create saltwater, salt is dissolved into water.

Solute: salt

Solvent: water

Semipermeable membrane: A membrane or barrier that allows only certain molecules or substances to pass through it and does not allow others.

Ex: Dialysis tubing acts as a semipermeable membrane.

Lugol's iodine: a solution of potassium iodide that turns blue in the presence of starch.

Diffusion Osmosis Lab Manual Answers

**Dr. Priyanka Gupta, Mr. Kaushalesh
Dwivedi**



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