

# **STUDY GUIDE FOR CHEMISTRY REVISION**

## **Chemical Reactions**

### **Question 1: Reactions and Equations**

For each statement, write an equation or chemical reaction to illustrate it, or if there is no reaction write chemical symbols.

- \_\_\_\_\_ a. an element reacts with oxygen
- \_\_\_\_\_ b. an element reacts with a non-metal element
- \_\_\_\_\_ c. an element reacts with water
- \_\_\_\_\_ d. a metal reacts with a non-metal element to form a compound
- \_\_\_\_\_ e. a metal reacts with an acid
- \_\_\_\_\_ f. a metal reacts with a salt solution
- \_\_\_\_\_ g. a compound reacts with an acid to form a salt and water
- \_\_\_\_\_ h. a compound reacts with a base to form a salt and water

Write the balanced chemical equation for each reaction. Write the names of the reactants, products, and any other substances.

reactants	products	gas	aq
balanced	balanced	gas	aq

Write the balanced chemical equation for the reaction of hydrogen gas with oxygen gas to form water. Write the names of the reactants, products, and any other substances.

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# Ch 9 Chemical Reactions Study Guide Answers

**Seager/Slabaugh**



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