

Chapter Test B

A. Matching

Match each item in Column A with the correct description in Column B. Write the letter of the correct description in the blank.

Column A

1. An unknown mass dissolved in a known volume of solvent for solution formation and analysis
2. An unknown volume of gas produced from a known mass of reactant
3. The relationship of quantities in chemical equations
4. The ratio of the actual yield to the theoretical yield expressed as a percent
5. The unknown mass of a product or reactant in a chemical reaction
6. An unknown amount of substance that reacts with a known mass of another substance

Column B

- a. percent yield
- b. limiting reagent
- c. theoretical yield
- d. stoichiometry
- e. percent yield
- f. percent reagent

B. Multiple Choice

Choose the best answer for each question from the list.

1. Which of the following is correct?
a. moles of reactants
b. moles of products
c. atoms of reactants
d. both reactants and products
2. Which of the following is correct?
a. moles of reactants = moles of products
b. atoms of reactants = atoms of products
c. mass of reactants = mass of products
d. all of the above
3. Which of the following is correct?
a. mass of reactants = mass of products
b. mass of reactants = mass of products
c. mass of reactants = mass of products
d. mass of reactants = mass of products
4. Which of the following is correct?
a. mass of reactants = mass of products
b. mass of reactants = mass of products
c. mass of reactants = mass of products
d. mass of reactants = mass of products

Ch 12 Stoichiometry Test B Pearson

**Theodore L. Brown, H. Eugene LeMay
Jr., Bruce E. Bursten, Catherine
Murphy, Patrick Woodward, Steven
Langford, Dalius Sagatys, Adrian
George**

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Prentice Hall Chemistry Antony C. Wilbraham, 2008 **Chemistry: The Central Science** Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Darius Sagatys, Adrian George, 2013-10-04

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