

## **Chemistry Formulas**



Dilution

Energy of a photon

Pressure

Average Kinetic Energy

Ideal Gas Law

Combined Gas Law

Dalton's Law of Partial

Pressures.

Dalton's Law of Partial

Pressures.

Graham's Law of Effusion

Henry's Law

Freezing Point Depression

Boiling Point Elevation

Vapor Pressure Depression (Rapult's Law)

Osmotic Pressure (n)

Arrhenius Equation

Equilibrium Constant Expressions

Equilibrium Constant Expinessions:

Equilibrium Constant Expressions

Reaction Quotient (Q)

Solubility Product Constant (K ...)

HOQ & HQ

HOQ & HQ

pH & pOH

M.V. = M.V.

 $E_{photon} = hf = hc/y$ 

P = F/A

 $KE_{max} = \frac{3}{2}RT$ 

PV = mRT

 $\frac{P_1V_1}{N_1T_1} = \frac{P_2V_2}{N_2T_2}$ 

P total = PA + Pa+ ......

 $P_A = \chi A P_{\text{notat}}$ 

 $\frac{T_{\text{min}}}{T_{\text{min}}} \sqrt{\frac{M_{\text{min}}}{M}}$ 

 $P_A = k_B E A T$ 

 $\Delta T_{\pi} = -iK_{\pi}M$ 

 $\Delta T_{B} = iK_{B}m$ 

Posts = Xanto Popular

m = iMRT

 $k = Ae^{-\epsilon \omega m \pi}$ 

 $K_{x} = \frac{[products]}{[reacturis]}$ 

 $K_{eq} = \frac{k_{\text{comman}}}{k}$ 

 $K_P = \frac{P_{maximin}}{P_{maximin}}$ 

 $Q = \frac{[produstets]}{[resotants]}$ 

 $K_{sp} = \frac{[products]}{[reactants]}$ 

 $pH = -\log TH^{-r}$ 

 $pOH = -\log[OH]$ 

pH + pOH = 1+

pK. & pK.

pK, & pK, pK, & pK,

Neutralization Reaction

Bufflers

Enthaloy of Formation

First Law of Thermodynamics

Pressure Volume Work

Calorimetry Thermal Energy

(p)

Heat Curves & Thermal Energy (o)

Heat Curves & Thermal Energy (q)

Heat Curves & Thermal Energy (q)

Entropy (S)

**Bond Dissociation Energy** 

Gibb's Free Energy (AG)

Gibb's Free Energy (AG)

Gibb's Free Energy (AG)

Standard Cell Potential

Standard Cell Potential

Nernst Equation

Kinetics (always 1st order)

Kinetics (always 1st order)

Kinetics (always 1st order)

Nuclear Binding Energy

 $pK_{-} = -\log[K_{-}]$ 

 $pK_{*} = -\log[K_{*}]$ 

 $pK_a + pK_b = 14$ 

 $n_a M_a V_a = n_a M_a V_a$ 

 $pH = pKa + \log [A - 1]HA1$ 

 $\Delta H_{\tau} = En\Delta H^{+}_{\tau}(product) -$ 

EndH' (reactants)

 $\Delta E = q + w$ 

W = -PAV

 $q = -C_{colorimotor}\Delta T$ 

 $q = mc\Delta T$ 

 $q = m\Delta H_{funter}$ 

 $q = m\Delta H_{magnification}$ 

 $\Delta S = EnS_{products} - EnS_{reactents}$ 

 $\Delta H = \Sigma \Delta H_{exactance} -$ 

 $\Delta G^* = \Delta H^* - T\Delta S^*$ 

EAH made to

 $\Delta G = \Delta G^* + RTInO$ 

 $\Delta G^* = -RT^*ln(K_{--})$ 

 $E^* = E^* reduction +$ 

E"oxidation

 $E^* = E^* cathode + E^* anode$ 

 $E_{cell} = E^* - \frac{0.0592}{2} \log \Omega$ 

 $M = M_{\pi} e^{-kx}$ .

 $\ln N = \ln N_n - kt$ 

 $t_{1/2} = \frac{0.693}{5}$ 

 $E = A \operatorname{cmc}^2$ 

# **Ch 12 Guide Chemical Calculations**

**C Cleary** 

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