

1. Choose the correct answer

a. The nitrate salt which does not give a mixture of NO_2 and O_2 on heating is

- (i) AgNO_3
- (ii) KNO_3
- (iii) $\text{Cu}(\text{NO}_3)_2$
- (iv) $\text{Zn}(\text{NO}_3)_2$

b. The chemical used in brown ring test is

- (i) CuSO_4
- (ii) FeSO_4
- (iii) $\text{Fe}_2(\text{SO}_4)_3$
- (iv) ZnSO_4

c. Lead nitrate decomposes on heating to give

- (i) NO
- (ii) N_2O
- (iii) NO_2
- (iv) N_2O_5

Solution:

- a. KNO_3
- b. FeSO_4
- c. NO_2

2. Name:

- a. A nitrate on heating which on heating does not give nitrogen dioxide
- b. A nitrate which on heating leaves no residue.
- c. A metal nitrate which on heating is changed to metal oxide.
- d. A metal nitrate which on heating is changed to metal.
- e. A solution which absorbs nitric oxide.
- f. The oxide of nitrogen which turns brown on exposure to air. How is it prepared?

Solution:

- a. Sodium nitrate
- b. Ammonium nitrate
- c. Calcium nitrate
- d. Silver nitrate
- e. Freshly prepared ferrous sulphate
- f. Nitric oxide

It is prepared by the catalytic oxidation of ammonia.



3. Mention three important uses of nitric acid. Give the property of nitric acid involved in the use.

Solution:

The uses of nitric acid along with the property involved include:

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