CLASS SET!! DO NOT TAKE!! DO NOT WRITE ON!!

Alchemy Unit Concepts Review

 Create charts similar to this one and fill in the similarities and differences for the bond types. Use your brain!! There is something you can put in every box!

Ionic	Covalent	Metallic
A		
2		
List 5 Examples	List 5 Examples	List 5 Examples

- What is the difference between core electrons and valence electrons? Calculate the number of core and valence electrons for O, Mg, Sn, and Pb.
- 3. Write the electron configurations for the following elements:

1000	THE CHOCKEO	ar connigurations	TANK THE	increase, and its reality	ACID CHARGOTTON	30.
a.	O				d.	1
b.	Ca				e.	5
C.	Mo					

4. What is an isotope? Predict the different isotopes for the following elements and write your answer in isotope symbols. Circle the isotope that is found in more abundance.

a.	Li	c.	
b.	Sr	d.	St

- On your paper, sketch an outline of the periodic table. On your sketch, label the following groups and trends:
 - a. Noble Gases
 b. Alkali Earth metals
 c. Transition metals
 d. Halogens
 e. Alkali metals
 f. Metals vs. nonmetals (where does
 - Metals vs. nonmetals (where does
 H belong?)

 Trend for atomic number (both)
 - g. Trend for atomic number (both across a period and down a group)
- across a period and down a group)

 6. In the periodic table, in general which elements will have similar properties?

h. Trend for atomic radius (size) (both across a period and down a group) i. Trend for reactivity down a group j. Trend for softness of metals down a group

Trend for making similar compounds in a group

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Concepts Review Unit D Answers

Michael Rajnik



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