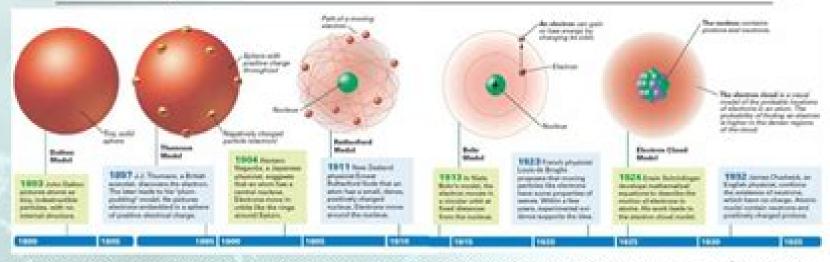
History of Modern Atomic Theory



- Niels Bohr noticed that electrons could only gain certain amounts of energy
 - This implied that there were energy levels in the electron cloud
 - Electrons could jump from one energy level to another, but could not exist between levels. The levels were still pictured as circular orbits.
 - This model of the atom was called the Bohr model and agree with experimental data for hydrogen atoms. It failed to explain other atoms though.

Concept Review Modern Atomic Theory

Robert C. Smoot, Smoot, Richard G. Smith, Jack Price

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