

Concentration Practice Problems

% Solution

1. What is the concentration (% solution) of a solution that contains 0.78 grams of MgCl_2 dissolved in 10. grams of water?

$$\% = \frac{\text{mass of solute}}{\text{mass of sol'n}} = \frac{0.78 \text{ g}}{10.78 \text{ g}} = 7.24\%$$

2. How many grams of solid are needed to prepare a 10.0% NaCl solution in 200. grams of water?

$$0.10 = \frac{x}{x+200}$$

$$0.10(x+200) = x$$

$$\begin{array}{r} 0.10x + 20 = x \\ -0.10x \quad \quad -0.10x \\ \hline 20 = 0.9x \end{array}$$

$$\frac{20}{0.9} = \frac{0.9x}{0.9}$$

$$x = 22.2 \text{ g NaCl}$$

3. What is the % solution of a mixture that contains 0.0500 moles of NaI dissolved in 25.0 g of water?

$$\frac{0.0500 \text{ moles NaI}}{1} \times \frac{150.0 \text{ g}}{1 \text{ mole}} = 7.5 \text{ g NaI}$$

$$\frac{7.5 \text{ g NaI}}{(7.5+25) \text{ g Sol'n}} = 23.1\%$$

Molarity

4. What is the molarity of a solution that contains 85.0 g of iodine (I_2) dissolved in 250. mL of ethanol solution?

$$\frac{85.0 \text{ g}}{1} \times \frac{1 \text{ mole}}{254 \text{ g}} = 0.335 \text{ mole}$$

$$M = \frac{0.335 \text{ moles}}{0.250 \text{ L}} = 1.34 \text{ M}$$

5. How many grams of $\text{Ca}(\text{NO}_3)_2$ are need to produce 1500. mL of a 0.75 M solution?

$$M = \frac{\text{mol}}{\text{L}}$$

$$0.75 = \frac{x}{1.500 \text{ L}}$$

$$x = 1.125 \text{ mol}$$

$$\frac{1.125 \text{ mol}}{1} \times \frac{164.1 \text{ g}}{1 \text{ mol}} = 184.6 \text{ g Ca}(\text{NO}_3)_2$$

Concentration Practice Answers Skills 39

JA Banks



Concentration Practice Answers Skills 39:

Concentration Practice Answers Skills 39 Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Concentration Practice Answers Skills 39**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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