Skills Worksheet

2.2 Concept Review

Section: Properties of Matter

1. Classify each of the following as a physical or chemical property of sulfur.

a. Its density is 2.97 g/cm³.

b. It reacts with hydrogen to form a gas.

c. It is a yellow solid.

d. Its melting point is 112°C.

e. It combines with oxygen.

Classify each of the following as a physical or chemical property of phosphorus.

a. It is a white, waxy solid.

b. It burns in air.

c. Its melting point is 44.1°C.

d. It has a density of 1.82 g/cm³.

c. Its boiling point is 280.3°C.

- Explain why aluminum is a suitable material to use in making cans based on its physical and chemical properties.
- Calculate the mass of a sample of pure silver (density = 10.49 g/cm³) that has a volume of 12.99 cm³.
- Compute the density of an 820 g sample of pure silicon occupying a 350 cm³ container.

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RS Peters

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