

Calorimetry Guided Notes

HS-PS3-4

CALORIMETRY GUIDED NOTES

Introduction to Calorimetry

Objectives

Students will understand the relationship between heat and temperature. They will be able to calculate the heat of reaction for a chemical reaction. They will be able to calculate the heat of formation for a compound. They will be able to calculate the heat of combustion for a fuel.

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Name _____ Date _____ Period _____

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What is Calorimetry?

Calorimetry is the study of heat. It is the measurement of the heat of a chemical reaction or physical process. It is the measurement of the heat of a chemical reaction or physical process.

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Heat & Temperature vs. Calorimetry

Calorimetry is the study of heat. It is the measurement of the heat of a chemical reaction or physical process. It is the measurement of the heat of a chemical reaction or physical process.

1. Heat is the energy that is transferred from one object to another. It is the energy that is transferred from one object to another. It is the energy that is transferred from one object to another.

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