

CP CHEMISTRY Periodic Table Review

Answer Key

1. Development of the Periodic Table

Scientist	Contribution to Periodic Table
John Newland	Arranged elements by increasing atomic mass, found repeating pattern every 8 elements, law of octaves
Dmitri Mendeleev	Arranged elements by increasing atomic mass, made accurate predictions about properties of missing elements, published work first – so gets credit for developing the first periodic table
Henry Moseley	Arranged elements by increasing atomic number which corrected problems with Mendeleev's table (elements out of order based on properties), his revisions lead to periodic law

2. What is the periodic law?

A periodic repetition of chemical and physical properties of the elements is observed when they are arranged by increasing atomic number.

3. What are the columns in the periodic table called? groups/families

4. What are the horizontal rows called? periods

5. What are the representative elements? Groups 1A to 8A

6. What are the transition elements? all Group B elements

7. What are the properties of the metals?

shiny, good conductors, solid at room temperature, ductile and malleable
most elements are metals

8. What are the metalloids?

found on stair step line (except Al), have chemical and physical properties of both metals and nonmetals

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Organic Chemistry Study Guide Robert J. Ouellette, J. David Rawn, 2014-11-04 Organic Chemistry Study Guide Key Concepts Problems and Solutions features hundreds of problems from the companion book Organic Chemistry and includes solutions for every problem Key concept summaries reinforce critical material from the primary book and enhance mastery of this complex subject Organic chemistry is a constantly evolving field that has great relevance for all scientists not just chemists For chemical engineers understanding the properties of organic molecules and how reactions occur is critically important to understanding the processes in an industrial plant For biologists and health professionals it is essential because nearly all of biochemistry springs from organic chemistry Additionally all scientists can benefit from improved critical thinking and problem solving skills that are developed from the study of organic chemistry Organic chemistry like any skill is best learned by doing It is difficult to learn by rote memorization and true understanding comes only from concentrated reading and working as many problems as possible In fact problem sets are the best way to ensure that concepts are not only well understood but can also be applied to real world problems in the work place Helps readers learn to categorize analyze and solve organic chemistry problems at all levels of difficulty Hundreds of fully worked practice problems all with solutions Key concept summaries for every chapter reinforces core content from the companion book *Study Guide [to Accompany]*

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