# CSP Exam Practice Test 2

- Q1. In the workplace, which specific function of the skin holds particular relevance for safeguarding against exposure to hazardous materials in the context of safety assessments?
- A. Sensory perception
- B. Immune function
- C. Fluid balance.
- D. Regulation of internal processes
- Q2. Elucidate the relationship between atomic number, electron configuration, and chemical reactivity in the formation of compounds, considering the periodic table.
- Atomic number directly correlates with chemical reactivity.
- B. Electron configuration is unrelated to atomic number.
- C. Chemical reactivity is solely determined by electron configuration.
- D. Atomic number and electron configuration collectively influence chemical reactivity.
- Q3. How do specific factors relate to the electronegativity of the atoms involved and how do these specific factors influence the strength of a covalent bond in chemical bonding?
- A. Molecular weight and atomic radius; inversely proportional
- Electronegativity difference and number of shared electrons; directly proportional
- C. Ionization energy and atomic mass; inversely proportional
- D. Atomic number and bond length; directly proportional
- Q4. How does the ionized nature of plasma, as the fourth state of matter pose unique

# **Csp Exam Study Guide**

**Jin-Ying Zhang** 

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