Electron	Configuration	Practice
Chemistr	y	

Maime:		
Dans Da		

How to write an electron configuration:

- A. Determine the total number of electrons to be represented.
- B. Use the Aufbau principle to fill the orbitals with electrons for elements 1-23. Refer to electron configuration periodic table for elements after 23
- C. The sum of the superscripts should equal the total number of electrons. Example: 12Mg Is² 2s² 2p⁶ 3s²

I. Configuration Writing Practice

- A. Write a ground state electron configuration for each neutral at: 1. Ground state means that all of the lowest possible energy levels (up to the pr per number of electrons for the element) are filled.
- 1. No: 1522522p6351
- 2. Pb: 1522522p63523p64523d104p65524d105p66524f145d106p2
- 3. Sr: 1522522p63523p64523d104p6552
- 4. U: 1522522p63523p64523d104p65524d105p66524f145d106p67525f36d1
- 5. N: 152 252 2p3
- *6. Ag: 1522522p63523p64523d104p65514d10
 - 7. Ti: 1522522p6 3523p6 452 3d2
 - 8. Ce: 1522522p63623p64523d104p65524d105p66524f15d1
 - 9. cl: 1522522p63523p5
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Neil D. Jespersen, Alison Hyslop

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also strives to provide actionable insights By shedding light on the challenges faced by marginalized groups the research aims to inform policy interventions and social initiatives that can address and mitigate the adverse effects of systemic exclusion Through this lens the study contributes to ongoing discussions on social justice equity and inclusive development in the Southeast Asian context fostering a deeper understanding of these issues for the benefit of both academic discourse and practical implementation **Chemistry** Theodore Lawrence Brown,1988

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