

## (Chapter 10: Thermal Energy and Heat Transfer: Caloric)

1. If heat transfer is perfectly insulated object is an adiabatic system.
2. A process is the transformation from one (initial) state to another.
3. The amount of energy required to raise the temperature of a kilogram of a substance by 1 kelvin is called its specific heat.
4. As the thermal energy of matter increases, its particles usually spend less time being stationary.
5. Temperature can be defined from the fact it is related to heat.
6. The property of thermal energy that is related to the number of particles (atoms or molecules) is called heat capacity.
7. The transfer of energy by electromagnetic waves is called radiation.
8. If heat is transferred between two objects, the amount of heat transferred is proportional to the mass of the objects and the change in temperature.
9. Suppose we have good insulation between the two objects.
10. If the heat capacity of the objects is the same, the amount of heat transferred is the same.
11. As the heat capacity of the objects increases, the amount of heat transferred is proportional to the change in temperature.
12. If the heat capacity of the objects is the same, the amount of heat transferred is the same.
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# Energy And Heat Study Guide Answers

**Peter L. Kresan, Reed Mencke, Frank  
Press, Raymond Siever**



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