Accuracy, Precision & Errors

(Quia quiz: http://www.guia.com/guiz/3507626.html)

There is always an element of error in our measurements.

The **accuracy** of a measurement is how <u>close it lies to the true value</u> ("correct answer"). The **precision** of a measurement is how <u>repeatable</u> that measurement is.



- High precision and high accuracy? Top Middle Bottom
- Low precision and low accuracy? Top Middle Bottom
- High precision but low accuracy? Top Middle Bottom

Systematic error is an error that is the same for all measurements and can be adjusted. For example, a digital balance that is poorly calibrated might measure 0.5g too much each time.

Random error is not consistent between measurements and cannot be adjusted. For example, errors introduced by using different recording tools or people.



Which of the dartboards shows:

4. Random error? Top - Middle - Bottom 5. Systematic error? Top - Middle - Bottom

Dartboard diagrams from http://preparatoryshemistry.com/Bishop Book 1 eBook.pdf

Measuring Temperature

Compare the glass thermometer with the temperature probe on the data logger.

Remember: Thermometers and dataloggers are NOT for stirring!

- ☐ Measure the temperature of the solution/ liquid NOT the container
- Don't touch the thermometer or probe against the glass
- □ Wait for the temperature to stabilize before taking the reading

Glass thermometer °C (±)	
Temperature probe °C (±)	25

Which would you choose to get quick readings of a reaction that changes temperature?

What is a Solution?

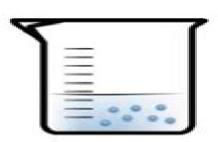
A solution is a homogenous* mixture of a substance in a liquid.

- The solvent is the liquid, e.g. water or alcohol, which dissolves the...
- Solute, which is the solid, e.g. salt or sugar.
- Label the solvent and the solute in the diagram to the right.

*mixtures can be:

- Homogenous = evenly distributed, like solutions
- Heterogeneous = unevenly distributed, like soup or the oceans.

Beaker from: http://www.wpclipart.com/science/beaker/beaker.png.html



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