

Chapter 15 Equilibrium Problems

Section 15.1 – 15.2

1. Write the equilibrium constant expression for K_c for the following reactions:



2. Write the equilibrium constant expression for K_p for $\text{H}_2\text{(g)} + \text{Cl}_2\text{(g)} \rightleftharpoons 2\text{HCl(g)}$

3. Write the chemical equation for ammonia formation and hydrogen, K_c is 0.50 at 400°C.



Calculate K_p for this reaction at the temperature

4. If at equilibrium $2\text{NO}_2\text{(g)} \rightleftharpoons \text{N}_2\text{(g)} + 2\text{O}_2\text{(g)}$ at a temperature of 400°C, the mass of oxygen at 0.050 mol O_2 . Calculate the value of K_c .

5. Write the equilibrium constant expressions for the acid dissociation of the following acids:



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Sabine Zange



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