Exercise 1: Esterification

Test Tube	Reaction Mixture CH ₃ COOH + octanol + H ₂ SO ₄	Aroma observed before and after reaction	Structure and IUPAC name of product formed		
A			THE RESERVE OF THE PARTY OF THE		
В	CH ₃ COOH + pentanol + H ₂ SO ₄	The same of the sa			
C	salicylic acid + methanol + H ₂ SO ₄				

Questions:

1.	Write a balanced cl	hemical equation	for each	reaction mi	xture showing	chemical:	structures.
	3.						



e.

 If you mixed 10 drops of ethanol and 10 drops of CH₃COOH, calculate the theoretical mass in grams of the ester formed in the reaction. Show all calculation steps. Assume that 20 drops = 1mL; density of ethanol=0.90 g/mL; [CH₃COOH]=17.4 M.

Esterification Experiment Grade 1

Philip H. Howard

Esterification Experiment Grade 1:

Biofuel and Bioenergy Technology Wei-Hsin Chen, Keat Teong Lee, Hwai Chyuan Ong, 2019-03-14 The subject of this book is Biofuel and Bioenergy Technology It aims to publish high quality review and research papers addressing recent advances in biofuel and bioenergy State of the art studies of advanced techniques of biorefinery for biofuel production are also included Research involving experimental studies recent developments and novel and emerging technologies in this field are covered This book contains twenty seven technical papers which cover diversified biofuel and bioenergy technology related research that have shown critical results and contributed significant findings to the fields of biomass processing pyrolysis bio oil and its emulsification transesterification and biodiesel gasification and syngas fermentation and biogas methane bioethanol and alcohol based fuels solid fuel and biochar and microbial fuel cell and power generation development The published contents relate to the most important techniques and analyses applied in the biofuel and bioenergy technology

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