

Physical Science – 2nd Semester – Final Exam Study Guide – (2012–2013)

1. Physical Science Basics

- What tool(s) would you want to use to find the:
 - Mass of an object? triple beam balance
 - Volume of an object? ruler or water displacement method
 - Length of an object? Ruler
 - Density of an object? graduated cylinder
- What is the formula for calculating density? mass/volume
 - What is the density of a cube that is 25 g and has a volume of 3 cm³? 8.33/cm³
 - Compare the density of solid H₂O to liquid H₂O, and explain what observable effect this has.
solid H₂O less dense than liquid H₂O

2. Chemistry Basics

- Element (def.) substance made up of one kind of atom
- Atom (def.) basic building block of matter
- Compound (def.) a molecule that contains atoms or molecules of two or more different elements
- Molecule (def.) a combination of two or more of the same or different atoms that are bonded together

3. Model of the Atomic Evolution of the Atomic Theory

Atoms are so small that many scientists throughout history have created models to describe them.

Why have atomic theories changed over time? Better technology came around.

Scientist	Year	Major Discovery/ Contribution	Name of atomic model
John Dalton	1808	Each element is made up of small, indivisible particles called atoms. Different elements have atoms of different masses.	Billiard Ball Model
J.J. Thomson	1897	Negatively charged electron	Plum Pudding Model
Niels Bohr	1913	Electrons move <u>fixed orbits</u> around the atom's nucleus in specific layers, or <u>electron shells</u> . Atoms absorb or give off energy when the electrons move from one shell (<u>energy level</u>) to another.	Bohr Model
Ernest Rutherford	1909	1909 /Based on idea that positive reflects positive/ atom mostly empty space; electrons orbit randomly around positively charged nucleus	Planetary Model

4. Periodic Table

- The current periodic table is arranged by increasing atomic number.
- Reading the Periodic Table
 - An element's properties can be predicted from its location in the periodic table.
 - The atomic number represents the number of protons.
 - The atomic mass represents the number of protons and neutrons.

Final Exam Study Guide Physical Science

Xiang Xie



Final Exam Study Guide Physical Science:

The book delves into Final Exam Study Guide Physical Science. Final Exam Study Guide Physical Science is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Final Exam Study Guide Physical Science, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Final Exam Study Guide Physical Science
 - Chapter 2: Essential Elements of Final Exam Study Guide Physical Science
 - Chapter 3: Final Exam Study Guide Physical Science in Everyday Life
 - Chapter 4: Final Exam Study Guide Physical Science in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Final Exam Study Guide Physical Science. The first chapter will explore what Final Exam Study Guide Physical Science is, why Final Exam Study Guide Physical Science is vital, and how to effectively learn about Final Exam Study Guide Physical Science.
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 4. In chapter 3, the author will examine the practical applications of Final Exam Study Guide Physical Science in daily life. The third chapter will showcase real-world examples of how Final Exam Study Guide Physical Science can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of Final Exam Study Guide Physical Science in specific contexts. This chapter will explore how Final Exam Study Guide Physical Science is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about Final Exam Study Guide Physical Science. The final chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Final Exam Study Guide Physical Science.

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