

CLASS SET!! DO NOT TAKE!! DO NOT WRITE ON!!

Alchemy Unit Concepts Review

1. Create charts similar to this one and fill in the similarities and differences for the bond types. Use your brain!! There is something you can put in every box!

Ionic	Covalent	Metallic
A		
2		
List 5 Examples	List 5 Examples	List 5 Examples

2. What is the difference between core electrons and valence electrons? Calculate the number of core and valence electrons for O, Mg, Sn, and Pb.
3. Write the electron configurations for the following elements:
- | | |
|-------|------|
| a. O | d. I |
| b. Ca | e. S |
| c. Mo | |
4. What is an isotope? Predict the different isotopes for the following elements and write your answer in isotope symbols. Circle the isotope that is found in more abundance.
- | | |
|-------|-------|
| a. Li | c. Si |
| b. Sr | d. Sb |
5. On your paper, sketch an outline of the periodic table. On your sketch, label the following groups and trends:
- | | |
|--|---|
| a. Noble Gases | h. Trend for atomic radius (size) (both across a period and down a group) |
| b. Alkali Earth metals | i. Trend for reactivity down a group |
| c. Transition metals | j. Trend for softness of metals down a group |
| d. Halogens | k. Trend for making similar compounds in a group |
| e. Alkali metals | |
| f. Metals vs. nonmetals (where does H belong?) | |
| g. Trend for atomic number (both across a period and down a group) | |
6. In the periodic table, in general which elements will have similar properties?

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Concepts Review Unit D Answers

K Payea



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