Skills Worksheet

2.2 Concept Review

Section: Properties of Matter

1. Classify each of the following as a physical or chemical property of sulfur.

a. Its density is 2.97 g/cm³.

b. It reacts with hydrogen to form a gas.

c. It is a yellow solid.

d. Its melting point is 112°C.

e. It combines with oxygen.

Classify each of the following as a physical or chemical property of phosphorus.

a. It is a white, waxy solid.

b. It burns in air.

c. Its melting point is 44.1°C.

d. It has a density of 1.82 g/cm³.

c. Its boiling point is 280.3°C.

- Explain why aluminum is a suitable material to use in making cans based on its physical and chemical properties.
- Calculate the mass of a sample of pure silver (density = 10.49 g/cm³) that has a volume of 12.99 cm³.
- Compute the density of an 820 g sample of pure silicon occupying a 350 cm³ container.

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Holt Chemistry Holt Rinehart & Winston, 2003-01-24 Concepts in Biology' 2007 Ed.2007 Edition, **Fbook:** Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature Matter, Building Block of the Universe ,1993 States of Matter, States of Mind Allan of Matter and Change F.M. Barton, 2021-08-29 States of Matter States of Mind is an easy to read introduction to the way the physical world is put together and stays together The book presents the fundamental ideas and particles of the makeup of the universe to enable understanding of matter and why it behaves in the way it does Written in an engaging manner the book explains some of the intricate details and grand schemes of life and the universe by making analogies with common everyday examples For example the recipe for a cake tells us nothing of how good the cake tastes but is a model of the food and a scientific model is no closer to the reality of the materials than a recipe is to the mouth watering flavor of the cake Illustrated with helpful cartoons this book provides a vast knowledge of atoms and atmospheres The first several chapters introduce terms and fundamental ideas while later chapters deal successively with particles and systems from the electron to the universe as a system Each new idea introduced builds upon the last A user friendly bibliography provides references for further reading

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