

## Skills Worksheet

## 2.2 Concept Review

### Section: Properties of Matter

1. **Classify** each of the following as a physical or chemical property of sulfur.

- \_\_\_\_\_ a. Its density is  $2.97 \text{ g/cm}^3$ .
- \_\_\_\_\_ b. It reacts with hydrogen to form a gas.
- \_\_\_\_\_ c. It is a yellow solid.
- \_\_\_\_\_ d. Its melting point is  $112^\circ\text{C}$ .
- \_\_\_\_\_ e. It combines with oxygen.

2. **Classify** each of the following as a physical or chemical property of phosphorus.

- \_\_\_\_\_ a. It is a white, waxy solid.
- \_\_\_\_\_ b. It burns in air.
- \_\_\_\_\_ c. Its melting point is  $44.1^\circ\text{C}$ .
- \_\_\_\_\_ d. It has a density of  $1.82 \text{ g/cm}^3$ .
- \_\_\_\_\_ e. Its boiling point is  $280.3^\circ\text{C}$ .

3. **Explain** why aluminum is a suitable material to use in making cans based on its physical and chemical properties.

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4. **Calculate** the mass of a sample of pure silver (density =  $10.49 \text{ g/cm}^3$ ) that has a volume of  $12.99 \text{ cm}^3$ .

5. **Compute** the density of an 820 g sample of pure silicon occupying a  $350 \text{ cm}^3$  container.

# Concept Review Section Energy Of State Changes

**Daniela Niemeyer**



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