

CHAPTER 1 THE LANGUAGE OF CHEMISTRY

Exercise 1(A)

Q.1. What is a symbol ? What is information does it convey?

Ans. Symbol :A symbol represent short form of an element.

It is convey following element.

- i) It represents a specific element.
- ii) It represents one atom of an element.
- iii) A symbol represents one atom of an element.
- iv) It represents no. of times an atom is heavier than one atomic mass unit (a.m.u) taken as a standard.

Q.2. Why is the symbol S for Sulphur, but Na for element represented by the Au,pb,Sn,Hg.

Ans. IUPAC means International Union of pure of and Applied Chemistry.

Au-Gold ,Pb-Lead, Sn-Tin, Hg – Mercury.

Q.4. IF the symbol for Cobalt, Co, were written a Co, what would be wrong with it ?

Ans. Co Stands for Cobalt, if we write it Co then it shows itself as a compound containing two non-metal i.e. Carbon and oxygen which form a gas carbon monoxide which is named by Greek Prefix.

Q.5. What do the following symbol for (a) H (b) H_2 (c) 2H (d) $2H_2$

Ans. (a)H stands for one atom of hydrogen.

(b) H_2 stands for one molecule of hydrogen.

(c) 2H Stands for one molecule of hydrogen.

(d) $2H_2$ Stands for two atoms and one molecule of hydrogen.

Q.6. What is meant by atomicity ? Name a diatomic element.

Ans. Atomicity : The number of atoms in a molecule of an element is called its atomicity.

Diatomic molecules – Element having two atoms in their molecules, e.g., hydrogen

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