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Solubility Curve Practice Problems Worksheet 1



You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any exceptions on the graph? _____

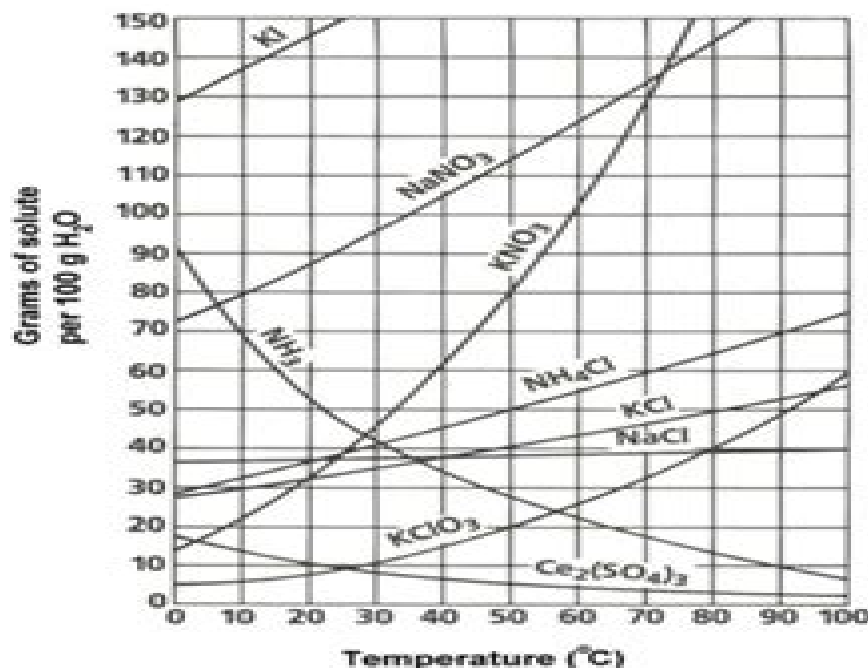
Here's an example of how to read the graph. Find the curve for KClO_3 .

At 30°C approximately 10g of KClO_3 will dissolve in 100g of water. If the temperature is increased to 80°C , approximately _____ of the substance will dissolve in 100g (or 100mL) of water.

Directions: Use the graph to answer the following questions. REMEMBER UNITS!

1) What mass of solute will dissolve in 100mL of water at the following temperatures?

- KNO_3 at 70°C = _____
- NaCl at 100°C = _____
- NH_4Cl at 90°C = _____
- Which of the above three substances is most soluble in water at 15°C . = _____



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Hussin A.Rothana

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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using materials Topics covered include the nature of materials such as crystals an atomic view of solids temperature effects on materials and the mechanical and chemical properties of materials This book is comprised of seven chapters and begins with an overview of the properties of different kinds of material the ways in which materials can be shaped and the uses to which they can be put The next chapter describes the state of matter as a balance between the tendencies of atoms to stick together by chemical bonding or rattle apart by thermal agitation paying particular attention to ionic bonds and ionic crystals the structure and properties of polymers and transition metals The reader is also introduced to how the structure of materials especially microstructure can be manipulated to give desired properties via thermal mechanical and chemical agents of change This text concludes by describing the chemistry of processing and service of various materials Exercises and self assessment questions with answers are given at the end of each chapter together with a set of objectives This monograph will be a valuable resource for students of materials science and the physical sciences

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